



# Structure Elucidation and Antimicrobial Activities of Secondary Metabolites from the Flowery Parts of *Verbascum mucronatum* Lam.

## *Verbascum mucronatum* Lam.'ın Çiçekli Kısımlarından Elde Edilen Sekonder Metabolitlerin Yapı Tayini ve Antimikrobiyal Aktiviteleri

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### ABSTRACT

**Objectives:** To determine the secondary metabolites from *Verbascum mucronatum* Lam. and evaluate their antimicrobial activity.

**Materials and Methods:** Antimicrobial activities of the isolated metabolites were determined using broth microdilutions against the bacteria (*Escherichia coli* ATCC 25922, *Enterococcus faecalis* ATCC 29212, *Pseudomonas aeruginosa* ATCC 27853, *Staphylococcus aureus* ATCC 29213) and fungi (*Candida albicans* ATCC 90028, *Candida krusei* ATCC 6258, *Candida parapsilosis* ATCC 90018).

**Results:** Four iridoid glycosides; ajugol (1), aucubin (2), lasianthoside I (3), catalpol (4), two triterpenic saponins; ilwensisaponin C (5), ilwensisaponin A (=mimengoside A) (6), and one phenylethanoid glycoside; verbascoside (=acteoside) (7) were isolated from the water soluble parts of the methanolic extract gained flowery parts of *V. mucronatum* Lam.

**Conclusion:** Within the obtained compounds, ajugol and ilwensisaponin A showed moderate antimicrobial activity, especially against fungi.

**Key words:** *Scrophulariaceae*, *Verbascum mucronatum* Lam., secondary metabolites, antimicrobial activity

### ÖZ

**Amaç:** Bu çalışmada *Verbascum mucronatum* Lam.'ın sekonder metabolitlerinin belirlenmesi ve antimikrobiyal aktivitelerinin değerlendirilmesi amaçlanmıştır.

**Gereç ve Yöntemler:** İzole edilen metabolitlerin, bakteri (*Escherichia coli* ATCC 25922, *Enterococcus faecalis* ATCC 29212, *Pseudomonas aeruginosa* ATCC 27853, *Staphylococcus aureus* ATCC 29213) ve mantarlara (*Candida albicans* ATCC 90028, *Candida krusei* ATCC 6258, *Candida parapsilosis* ATCC 90018) karşı antimikrobiyal aktiviteleri sıvı mikrodilüsyon yöntemiyle belirlenmiştir.

**Bulgular:** *V. mucronatum* Lam.'ın çiçekli kısımlarının metanol ektresinin suda çözünen kısımlarından, dört iridoit glikoziti, ajugol (1), okubin (2), lasiantozit I (3), katalpol (4); iki triterpenik saponin, ilwensisaponin C (5), ilwensisaponin A (=mimengozit A) (6) ve bir feniletanoit glikoziti, verbaskozit (=akteozit) (7) izole edilmiştir.

**Sonuç:** Elde edilen bileşikler içinde ajugol ve ilwensisaponin A, özellikle mantarlara karşı zayıf antimikrobiyal aktivite göstermiştir.

**Anahtar kelimeler:** *Scrophulariaceae*, *Verbascum mucronatum* Lam., sekonder metabolitler, antimikrobiyal aktivite

### INTRODUCTION

*Verbascum* is a widespread genus of the family *Scrophulariaceae*, which comprises more than 300 species of the world's flora.<sup>1</sup> This genus is represented by 233 species, 196 of which are endemic in Turkish flora.<sup>2-4</sup> Infusions prepared with the leaves

and flowers of *Verbascum* species have been used as an expectorant and mucolytic<sup>5</sup> wound healer<sup>6</sup> for the treatment of hemorrhoids and rheumatism<sup>7</sup> in folk medicine. Turker and Camper<sup>8</sup> showed that *Klebsiella pneumoniae* and *Staphylococcus aureus* showed sensitivity to Mullein (*Verbascum thapsus*), which may explain why Mullein is used in folk medicine to

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treat respiratory disorders (caused by *K. pneumoniae* and *S. aureus*) and urinary tract infections (caused by *K. pneumoniae*). Antibacterial and antifungal activities of *Verbascum* L. species have been previously reviewed and the activity of the genus against several bacteria and fungi has been revealed.<sup>9</sup> The antimicrobial activity of *Verbascum mucronatum* has also been determined using disc diffusion tests by our research group.<sup>10</sup> In addition, *V. mucronatum* Lam. has been used as a Hemostatic in Turkish traditional medicine.<sup>11</sup>

Previous investigations on Turkish *Verbascum* L. species by our research group led to the isolation and characterization of a number of secondary metabolites such as iridoids, monoterpene glucosides, saponins, phenylethanoids, neolignans, and flavonoid glycosides.<sup>12-16</sup> As a part of our ongoing studies on the secondary metabolites of *Verbascum* L. species, we have now investigated the methanolic extract of the flowery parts of *V. mucronatum*, and isolated four iridoids; ajugol (1), aucubin (2), lasianthoside I (3), catalpol (4), two saponins; ilwensisaponin C (5) and ilwensisaponin A (6), along with a phenylethanoid glycoside, verbascoside (=acteoside) (7) by means of various chromatographic techniques (Figure 1). The current paper deals with the isolation and structure elucidation of the compounds (1-7) from the title plant and the evaluation of their antimicrobial activities.

## MATERIALS AND METHODS

### General experimental procedures

The ultraviolet (UV) spectra ( $\lambda_{\max}$ ) were recorded on a Agilent 8453 spectrophotometer. The infrared (IR) spectra ( $\nu_{\max}$ ) were determined on a Perkin Elmer 2000 fourier transform (FT)-IR spectrophotometer. The 1D and 2D nuclear magnetic resonance (NMR) spectra were obtained on a Bruker Avance DRX 500 and 400 FT spectrometer operating at 500 and 400 MHz for <sup>1</sup>H NMR, and 125 and 100 MHz for <sup>13</sup>C NMR. For the <sup>13</sup>C NMR spectra, multiplicities were determined using distortionless enhancement with a polarization transfer (DEPT) experiment. LC-ESIMS data were obtained using a Bruker BioApex FT-mass spectrometry instrument in the ESI mode. Reversed-phase material (C-18, LiChroprep 25-40  $\mu$ m) and polyamide were used for vacuum liquid chromatography (VLC), reversed-phase material (C-18, LiChroprep 25-40  $\mu$ m) was used for middle pressure liquid chromatography (MPLC), and Si gel (230-400 mesh) (Merck) was used for column chromatography (CC). Pre-coated silica gel 60 F<sub>254</sub> aluminum sheets (Merck) were used for thin-layer chromatography (TLC); developing systems, CHCl<sub>3</sub>-MeOH-H<sub>2</sub>O (61:32:7 and 80:20:2). Plates were examined using UV fluorescence and sprayed with 1% vanillin in concentrated H<sub>2</sub>SO<sub>4</sub>, followed by heating at 105°C for 1-2 min.

### Plant material

*V. mucronatum* Lam. was collected from Aksaray, 17 km from Aksaray to Ulukışla, in July 2007. A voucher specimen has been deposited in the Herbarium of the Faculty of Science, Gazi University, Ankara, Turkey (GAZI 10097). The flowery parts of the plant, which were air dried in the shade, were used in the phytochemical studies.

### Extraction and isolation

Air-dried and powdered flowery parts of the plant (586.2 g) were extracted with MeOH (3x2.5 L). The MeOH extract was evaporated to dryness in vacuo to yield 70.4 g of crude extract, then MeOH extract was dissolved with 100 mL distilled water and partitioned in CHCl<sub>3</sub> (2x100 mL). H<sub>2</sub>O and CHCl<sub>3</sub> phases were evaporated to dryness in vacuo to yield 65.8 g H<sub>2</sub>O and 3.6 g CHCl<sub>3</sub> extracts. The H<sub>2</sub>O phase was fractionated using CC on polyamide (150 g) using H<sub>2</sub>O-MeOH (100:0→0:100) (each 500 mL), respectively, to yield 6 fractions (Frs. A-F). Fraction D (4.9 g), eluted with 75% methanol, was subjected to VLC using reversed-phase material (C-18, LiChroprep 25-40  $\mu$ m, 150 g), using MeOH-H<sub>2</sub>O mixtures (0-100%) to give catalpol (4) (62.1 mg), aucubin (2) (139.3 mg), ajugol (1) (48.6 mg), Fr. D3 (1.19 g) and Fr. D4 (625.3 mg). Frs. D3 and D4 were rechromatographed. Fr. D3 was applied to MPLC using reversed-phase material (C-18, LiChroprep 25-40  $\mu$ m) using MeOH-H<sub>2</sub>O mixtures (100:0→30-70) to yield ilwensisaponin C (5) (14.7 mg), ilwensisaponin A (6) (51.5 mg), and lasianthoside I (3) (6.7 mg). Fr. D4 was rechromatographed on a silica gel column (55 mg) and eluted CHCl<sub>3</sub>-MeOH (70:30→60:40) mixtures to give verbascoside (=acteoside) (7) (14.8 mg).

### Antimicrobial activity-broth microdilution method

Antibacterial and antifungal activities were determined using the broth microdilution test as recommended by Clinical and Laboratory Standards Institute.<sup>17,18</sup> Plant extracts were tested against four bacteria including two Gram-positive (*S. aureus* ATCC 29213, *Enterococcus faecalis* ATCC 29212) and two Gram-negative microorganisms (*Escherichia coli* ATCC 25922, *Pseudomonas aeruginosa* ATCC 27853), as well as for antifungal activities against three yeasts (*Candida albicans* ATCC 90028, *Candida krusei* ATCC 6258, *Candida parapsilosis* ATCC 90018). The antibacterial activity test was performed in Mueller-Hinton broth (MHB, Difco Laboratories, Detroit, MI, USA); for antifungal test, RPMI-1640 medium with L-glutamine (ICN-Flow, Aurora, OH, USA), buffered with MOPS buffer (ICN-Flow, Aurora, OH, USA) was used. The inoculum densities were approximately 5x10<sup>5</sup> CFU/mL and 0.5-2.5x10<sup>3</sup> CFU/mL for bacteria and fungi, respectively.

Each plant extract was dissolved in 2.44 mL DMSO. Finally, two-fold concentrations were prepared in the wells of the microtiter plates, between 1024-1  $\mu$ g/mL. Ampicillin and fluconazole were used as reference antibiotics for bacteria and fungi, respectively (64-0.0625  $\mu$ g/mL). Microtiter plates were incubated at 35°C for 18-24 h for bacteria and 48 h for fungi. After the incubation period, minimum inhibitory concentration (MIC) values were defined as the lowest concentration of the extracts that inhibits the visible growth of the microorganisms.

## RESULTS

**Ajugol (1):** UV  $\lambda_{\max}$  (MeOH) 220 nm, IR (KBr)  $\nu_{\max}$  3410 (OH), 1660 (C=C) cm<sup>-1</sup>, Positive ion LC-ESIMS m/z 371 [M+Na]<sup>+</sup> (calc. for C<sub>15</sub>H<sub>24</sub>O<sub>9</sub>), <sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>) of 1:  $\delta_{\text{H}}$  6.10 (1H, dd, *J*=6/1.6 Hz, H-3), 5.29 (1H, d, *J*=2 Hz, H-1), 4.78 (1H, dd, *J*=6/2.8 Hz, H-4), 4.43 (1H, d, *J*=7.6 Hz, H-1'), 3.71 (1H, d, *J*=2.8 Hz, H-6), 3.71-3.65 (2H, \*, H-6'), 3.05-2.93 (1H, \*, H-2', H-3', H-4', H-5'), 2.47 (1H, m, H-5), 2.32 (1H, t, *J*=10 Hz, H-9), 1.84 (1H,

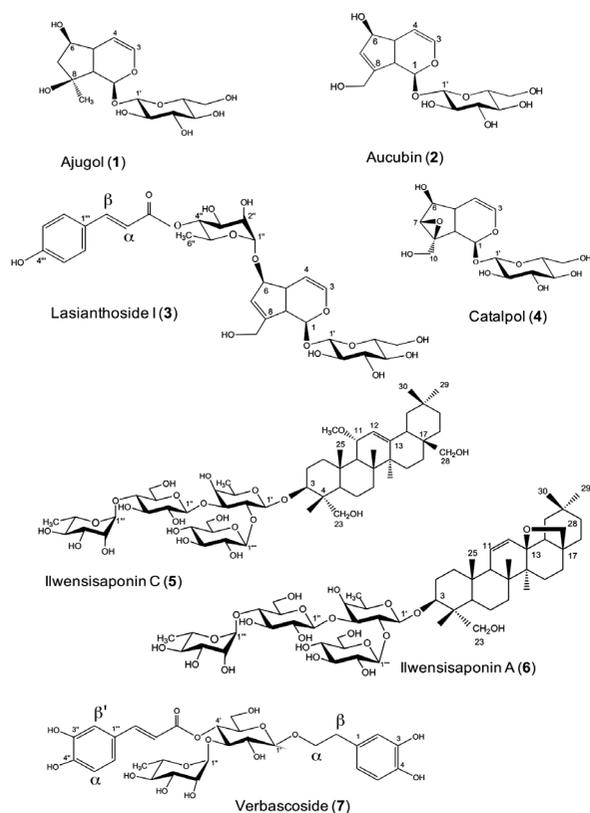


Figure 1. Isolated secondary metabolites from *Verbascum mucronatum* Lam.

Table 1.  $^{13}\text{C}$  NMR ( $\text{DMSO}-d_6$ ) data of compounds of 1, 2, 3 and 4

	2 (100 MHz)	3 (125 MHz)	4 (100 MHz)	1 (100 MHz)
C/H atom	$\delta_{\text{C}}$ (ppm)	$\delta_{\text{C}}$ (ppm)	$\delta_{\text{C}}$ (ppm)	$\delta_{\text{C}}$ (ppm)
Aglycone				
1	95.9	96.0	93.8	92.1
3	140.6	141.1	140.7	139.3
4	105.6	104.8	103.8	105.7
5	45.2	42.8	37.8	40.7
6	81.1	87.5	77.8	77.8
7	129.8	125.6	61.2	50.6
8	146.8	149.4	65.3	77.7
9	47.0	47.3	42.6	50.5
10	60.1	59.9	59.5	25.7
Glc at C-1				
1'	98.7	100.1	98.3	98.1
2'	74.0	73.9	73.8	73.8
3'	77.3	77.2	76.8	76.1
4'	70.7	70.7	70.6	70.7
5'	77.8	77.6	77.6	77.4
6'	61.7	61.6	61.7	61.7

Compound 3: Rha at C-6, 98.6 (C-1'), 74.3 (C4''), 71.4 (C-2''), 68.8 (C-3''), 67.0 (5''), 18.0 (C-6''); Acyl moiety, 166.8 (C=O), 161.0 (C-4'''), 145.2 (C- $\beta$ ), 133.0 (1'''), 130.7 (C-2'''), 130.7 (C-6'''), 116.3 (C-3'''), 116.3 (C-5'''), 115.4 (C- $\alpha$ )

dd,  $J=12.8/6.0$  Hz, H-7b), 1.63 (1H, dd,  $J=13.2/6.0$  Hz, H-7a), 1.13 (3H, s, H-10), and  $^{13}\text{C}$  NMR (100 MHz,  $\text{DMSO}-d_6$ ) (see Table 1).

**Aucubin (2):** UV  $\lambda_{\text{max}}$  (MeOH) 205 nm, (KBr)  $\nu_{\text{max}}$  3275 (OH), 1650 (C=C)  $\text{cm}^{-1}$ , Positive ion LC-ESIMS  $m/z$  369  $[\text{M}+\text{Na}]^+$  (calc. for  $\text{C}_{15}\text{H}_{22}\text{O}_9$ ),  $^1\text{H}$  NMR (400 MHz,  $\text{DMSO}-d_6$ ) of 2:  $\delta_{\text{H}}$  6.30 (1H, dd,  $J=4.8/1.6$  Hz, H-3), 5.65 (1H, bs, H-7) 5.01 (1H, d,  $J=4.8$  Hz, H-4), 4.95 (1H, d,  $J=5.6$  Hz, H-1), 4.85 (1H, d,  $J=7.7$  Hz, H-1'), 4.40, (1H, d,  $J=6.4$  Hz, H-6), 4.14 (1H, dd,  $J=12.4/4.0$  Hz, H-10b), 3.96 (1H, dd,  $J=12.4/4.0$  Hz, H-10a), 3.66 (1H, dd,  $J=12.8/4.8$  Hz, H-6'a), 3.42 (1H, dd,  $J=12.0/4.8$  Hz, H-6'b), 3.16 (1H, m, H-3'), 3.11 (1H, m, H-4'), 3.04 (1H, m, H-5'), 3.00 (1H, m, H-2'), 2.72 (1H, t,  $J=7.2$  Hz, H-9), 2.50 (1H, m, H-5), and  $^{13}\text{C}$  NMR (100 MHz,  $\text{DMSO}-d_6$ ) (see Table 1).

**Lasianthoside I (3):** UV  $\lambda_{\text{max}}$  (MeOH) 216, 277 nm, IR (KBr)  $\nu_{\text{max}}$  3405 (OH), 1704 (C=O), 1655 (C=C), 1508, 1451 (aromatic ring)  $\text{cm}^{-1}$ , Positive ion LC-ESIMS  $m/z$  611  $[\text{M}+\text{Na}]^+$  (calc. for  $\text{C}_{30}\text{H}_{38}\text{O}_{15}$ ),  $^1\text{H}$  NMR (400 MHz,  $\text{DMSO}-d_6$ ) of 3:  $\delta_{\text{H}}$  6.37 (1H, dd,  $J=4.8/1.2$  Hz, H-3), 5.26 (1H, d,  $J=4.4$  Hz, H-4), 5.10 (1H, d,  $J=4.0$  Hz, H-1), 4.91 (1H, d,  $J=7.6$  Hz, H-1'), 4.18 (1H, d,  $J=6.0$  Hz, H-10b), 3.86 (1H, d,  $J=4$  Hz, H-6'b), 3.78 (1H, t,  $J=6.8$  Hz, H-6), 3.66 (1H, \*, H-10a), 3.64 (1H, dd,  $J=10.8/6.4$  Hz, H-6'a), 3.35 (1H, s, H-7), 2.31 (1H, t,  $J=7.6$  Hz, H-9), 3.13-3.19 (1H, \*, H-3', H-4', H-5'), 3.02 (1H, dd,  $J=10/6.4$  Hz, H-2'), 2.12 (1H, m, H-5), and  $^{13}\text{C}$  NMR (125 MHz,  $\text{DMSO}-d_6$ ) (see Table 1).

**Catalpol (4):** UV  $\lambda_{\text{max}}$  (MeOH) nm 208 nm, IR (KBr)  $\nu_{\text{max}}$  3450 (OH), 1670 (C=C)  $\text{cm}^{-1}$ , Positive ion LC-ESIMS  $m/z$  385  $[\text{M}+\text{Na}]^+$  (calc. for  $\text{C}_{15}\text{H}_{22}\text{O}_{10}$ ),  $^1\text{H}$  NMR (400 MHz,  $\text{DMSO}-d_6$ ) of 4:  $\delta_{\text{H}}$  6.37 (1H, dd,  $J=4.8/1.2$  Hz, H-3), 5.26 (1H, d,  $J=4.4$  Hz, H-4), 5.10 (1H, d,  $J=4.0$  Hz, H-1), 4.91 (1H, d,  $J=7.6$  Hz, H-1'), 4.18 (1H, d,  $J=6.0$  Hz, H-10b), 3.86 (1H, d,  $J=4$  Hz, H-6'b), 3.78 (1H, t,  $J=6.8$  Hz, H-6), 3.66 (1H, \*, 10a), 3.64 (1H, dd,  $J=10.8/6.4$  Hz, H-6'a), 3.35 (1H, s, H-7), 3.13-3.19 (\*, H-3', H-4', H-5'), 3.02 (1H, dd,  $J=10/6.4$  Hz, H-2'), 2.31 (1H, t,  $J=7.6$  Hz, H-9), 2.12 (1H, m, H-5), and  $^{13}\text{C}$  NMR (100 MHz,  $\text{DMSO}-d_6$ ) (see Table 1).

**Ilwensisaponin C (5):** UV  $\lambda_{\text{max}}$  (MeOH) 205 nm, IR (KBr)  $\nu_{\text{max}}$  3400 (OH), 1665 (C=C)  $\text{cm}^{-1}$ , Positive ion LC-ESIMS  $m/z$  1127  $[\text{M}+\text{Na}]^+$  (calc. for  $\text{C}_{55}\text{H}_{92}\text{O}_{22}$ ),  $^1\text{H}$  NMR (400 MHz, pyridine) of 5:  $\delta_{\text{H}}$  5.78 (1H, bs, H-1'''), 5.54 (1H, d,  $J=7.0$  Hz, H-1'''), 5.46 (1H, bs, H-12), 5.21 (1H, d,  $J=7.0$  Hz, H-1''), 4.91 (1H, d,  $J=6.6$  Hz, H-1'), 4.35 (1H, \*, H-2'), 4.33 (1H, \*, H-23b), 4.10 (1H, \*, H-2'''), 4.10 (1H, \*, H-3), 3.89 (1H, \*, H-2''), 3.82 (1H, \*, H-11), 3.81 (1H, d,  $J=11.7$  Hz, H-28b), 3.69 (1H, d,  $J=8.3$  Hz, H-23a), 3.57 (1H, d,  $J=10.2$  Hz, H-28a), 1.68 (3H, d,  $J=5.5$  Hz, H-6'''), 1.35 (3H, d,  $J=4.8$  Hz, H-6'), 1.30 (3H, s, H-27), 1.08 (3H, s, H-24), 1.07 (3H, s, H-25), 0.96 (3H, s, H-26), 0.95 (3H, s, H-30), 0.88 (3H, s, H-29),  $\text{CH}_3\text{O}$ : 3.21 (3H, s), and  $^{13}\text{C}$  NMR (125 MHz, pyridine) (see Table 2).

**Ilwensisaponin A (6):** UV  $\lambda_{\text{max}}$  (MeOH) 206 nm, IR (KBr)  $\nu_{\text{max}}$  3434 (OH), 1645 (C=C)  $\text{cm}^{-1}$ , Positive ion LC-ESIMS  $m/z$  1095  $[\text{M}+\text{Na}]^+$  (calc. for  $\text{C}_{54}\text{H}_{88}\text{O}_{21}$ ),  $^1\text{H}$  NMR (500 MHz, pyridine) of 6:  $\delta_{\text{H}}$  5.94 (1H, d,  $J=10.4$  Hz, H-11), 5.77 (1H, d,  $J=1.5$  Hz, H-1''), 5.53 (1H, \*, H-12), 5.20 (1H, d,  $J=7.6$  Hz, H-1'), 5.53 (1H, d,  $J=7.9$  Hz, H-1'''), 4.91 (1H, d,  $J=7.7$  Hz, H-1'), 4.58 (1H, \*, H-2''), 4.34 (1H, \*, H-23b), 4.25 (1H, \*, H-2'), 4.11 (1H, \*, H-3), 4.05 (1H, \*, H-2'''), 3.90 (1H, \*, H-2''), 3.72 (1H, \*, H-28b), 3.70 (1H, \*, H-23a), 3.33

Table 2.  $^{13}\text{C}$  NMR (125 MHz, pyridine- $d_5/5$ ,  $\text{CD}_3\text{OD}/6$ ) data of compounds 5 and 6

	5	6		5	6
C/H atom	$\delta_{\text{C}}$ (ppm)	$\delta_{\text{C}}$ (ppm)	C/atom	$\delta_{\text{C}}$ (ppm)	$\delta_{\text{C}}$ (ppm)
Aglycone			Sugar units		
1	40.2	38.0	Fuc at C-3		
2	22.9	25.6	1'	104.2	104.7
3	83.0	84.0	2'	77.0	77.1
4	44.1	45.9	3'	85.0	85.7
5	48.1	46.0	4'	72.2	72.2
6	18.5	18.0	5'	70.6	70.7
7	31.9	31.0	6'	17.3	17.0
8	37.6	42.6	Glc at Fuc C-3'		
9	52.8	54.1	1''	105.1	105.1
10	35.8	37.0	2''	75.6	75.4
11	76.2	132.9	3''	77.8	76.1
12	122.6	131.9	4''	78.4	79.3
13	148.1	86.9	5''	77.2	76.4
14	43.6	44.1	6''	61.4	63.5
15	26.7	26.0	Rha at Glc C-4''		
16	26.4	26.5	1'''	102.8	102.9
17	42.2	40.0	2'''	72.8	72.7
18	42.5	52.8	3'''	72.6	71.3
19	47.1	38.3	4'''	74.0	73.8
20	31.4	31.0	5'''	70.5	70.7
21	33.3	34.0	6'''	18.5	18.5
22	34.8	32.0	Glc at Fuc C-2'		
23	64.8	64.5	1''''	104.0	103.5
24	13.4	12.6	2''''	76.2	75.4
25	18.0	19.0	3''''	78.8	76.8
26	18.7	22.0	4''''	72.2	73.5
27	26.4	20.0	5''''	76.5	78.3
28	68.9	78.3	6''''	63.3	61.8
29	33.5	34.0			
30	24.0	24.0			
$\text{OCH}_3$	54.1	-			

(1H, d,  $J=6.2$  Hz, H-28a), 1.68 (1H, d,  $J=6.1$  Hz, H-6'''), 1.38 (3H, bs, H-6'), 1.31 (3H, s, H-26), 1.04 (3H, s, H-24), 0.98 (3H, s, H-27), 0.96 (3H, s, H-25), 0.87 (3H, s, H-29), 0.82 (3H, s, H-30), and  $^{13}\text{C}$  NMR (125 MHz,  $\text{CD}_3\text{OD}$ ) (see Table 2).

**Verbascoside (Acteoside) (7):** UV  $\lambda_{\text{max}}$  (MeOH) 220, 332 nm, IR (KBr)  $\nu_{\text{max}}$  3392 (OH), 1699 (C=O), 1631 (C=C), 1604, 1525 (aromatic ring)  $\text{cm}^{-1}$ , Positive ion LC-ESIMS  $m/z$  647  $[\text{M}+\text{Na}]^+$  (calc. for  $\text{C}_{29}\text{H}_{36}\text{O}_{15}$ ),  $^1\text{H}$  NMR (500 MHz,  $\text{DMSO}-d_6$ ) of 7:  $\delta_{\text{H}}$  7.48 (1H, d,  $J=15.8$  Hz, H- $\beta'$ ), 7.04 (1H, s, H-2'''), 6.97 (1H, d,  $J=7.5$

Hz, H-6'''), 6.79 (1H, d,  $J=7.7$  Hz, H-5'''), 6.67 (1H, bs, H-2), 6.67 (1H, bs, H-5), 6.52 (1H, d,  $J=7.5$  Hz, H-6), 6.20 (1H, d,  $J=15.8$  Hz, H- $\alpha'$ ), 5.07 (1H, bs, H-1''), 4.75 (1H, t,  $J=9.4$  Hz, H-4'), 4.37 (1H, d,  $J=7.7$  Hz, H-1'), 3.72 (1H, \*, H-2''), 3.91, (1H, m, H- $\alpha_b$ ), 3.67, (1H, m, H- $\alpha_a$ ), 2.73 (2H, s, H- $\beta$ ), 3.68 (1H, \*, H-3'), 3.45-3.70 (2H, \*, H-6'), 3.45 (1H, \*, H-5'), 3.36 (1H, \*, H-5''), 3.35 (1H, \*, H-3''), 3.26 (1H, t,  $J=8.3$  Hz, H-2'), 3.15 (1H, \*, H-4''), 1.00 (3H, d,  $J=5.8$  Hz, H-6''), and  $^{13}\text{C}$  NMR (125 MHz,  $\text{CDCl}_3$ ) (see Table 3).

\*(overlapped)

The methanolic extract of the flowery part of *V. mucronatum* and isolated compounds possessed moderate antimicrobial activity, especially against fungi. Iridoid glycoside ajugol was found to be the most active compound against *C. albicans* and *C. parapsilosis* with an MIC value of 64 µg/mL, as well as ilwensisaponin A

inhibited *C. albicans* and *C. krusei* with the same MIC value as ajugol. These active compounds were found to be much more effective against fungi than the *V. mucronatum* extract (Table 4).

## DISCUSSION

Compound **1** was isolated as a white amorphous powder with the molecular formula  $C_{15}H_{24}O_9$  (LC-ESIMS  $m/z$  371 [M+Na]<sup>+</sup>). An iridoid enolether system (220 nm) in UV spectrum; hydroxyl group (3410  $cm^{-1}$ ) and double-bond (1660  $cm^{-1}$ ) absorption bands in IR spectra were observed. Compound **1** was identified as ajugol when comparing <sup>1</sup>H and <sup>13</sup>C NMR spectra with those of ajugol.<sup>19</sup>

Compound **2** (see Figure 1) was isolated as white amorphous powder with the molecular formula  $C_{15}H_{22}O_9$  (LC-ESIMS  $m/z$  369 [M+Na]<sup>+</sup>). An iridoid enolether system (205 nm) in UV spectrum; hydroxyl group (3275  $cm^{-1}$ ) and double-bond (1650  $cm^{-1}$ ) absorption bands in IR spectra were observed. Compound **2** was identified as aucubin when comparing <sup>1</sup>H and <sup>13</sup>C NMR spectra with those of aucubin.<sup>20,21</sup>

Compound **3** (see Figure 1) was isolated as a white amorphous powder with the molecular formula  $C_{30}H_{38}O_{15}$  (LC-ESIMS  $m/z$  661 [M+Na]<sup>+</sup>). The presence of an iridoid enolether system (216 nm) and an aromatic acid (277 nm) moiety in UV spectrum and absorption bands for a hydroxyl group (3405  $cm^{-1}$ ), a conjugated ester carbonyl (1704  $cm^{-1}$ ), a double-bond (1655  $cm^{-1}$ ) and an aromatic ring (1451  $cm^{-1}$ , 1508  $cm^{-1}$ ) in IR spectra were observed. The <sup>1</sup>H and <sup>13</sup>C NMR spectra of **3** were similar to those of lasianthoside I. Based on this evidence, compound **3** was identified as lasianthoside I.<sup>22</sup>

Compound **4** (Figure 1) was isolated as a white amorphous powder with the molecular formula  $C_{15}H_{22}O_{10}$  (LC-ESIMS  $m/z$  385 [M+Na]<sup>+</sup>). Its UV spectrum supported the presence of an iridoid enolether system (208 nm) and absorption bands were for a hydroxyl group (3450  $cm^{-1}$ ), and a double-bond (1670  $cm^{-1}$ )

**Table 3.** <sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>) data of compound **7**

7			
C/atom	δ <sub>c</sub> (ppm)	C/atom	
Aglycone		Rha at Glc C-3'	
1	131.5	1''	103.1
2	117.2	2''	72.3
3	146.7	3''	72.1
4	144.3	4''	73.9
5	116.7	5''	70.5
6	120.5	6''	18.9
α	71.4	Acyl moiety	
β	35.9	1'''	127.7
		2'''	115.6
Glc		3'''	146.9
1'	104.3	4'''	149.9
2'	76.3	5'''	116.4
3'	81.7	6'''	122.2
4'	70.7	α'	114.7
5'	76.1	β'	148.1
6'	62.7	C=O	168.3

**Table 4.** Minimum inhibitory concentrations (µg/mL) of the methanolic extract and the secondary metabolites

	Bacteria				Fungi		
	<i>Staphylococcus aureus</i>	<i>Enterococcus faecalis</i>	<i>Escherichia coli</i>	<i>Pseudomonas aeruginosa</i>	<i>Candida albicans</i>	<i>Candida krusei</i>	<i>Candida parapsilosis</i>
	ATCC 29213	ATCC 29212	ATCC 25922	ATCC 27853	ATCC 90028	ATCC 6258	ATCC 22019
<i>Verbascum mucronatum</i> -MeOH extract	256	128	256	256	256	128	128
Ajugol	128	256	128	128	64	128	64
Aucubin	256	512	512	256	128	256	256
Lasianthoside I	>512	512	512	512	256	512	256
Catalpol	256	512	512	256	256	256	256
Ilwensisaponin C	>512	>512	512	512	256	512	256
Ilwensisaponin A	256	>512	>512	512	64	64	128
Verbascoside	256	512	512	256	256	256	256
Ampicillin	1	8	2	-	-	-	-
Fluconazole	-	-	-	-	1	64	8

in the IR spectra were observed. The  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra of compound **4** were similar to those of catalpol. Thus, compound **4** was identified as catalpol.<sup>23</sup>

Compounds **5** and **6** (Figure 1) were obtained as amorphous compounds with molecular weights of 1104 {LC-ESIMS:  $m/z$  1127 ([M+Na]<sup>+</sup>)}, and 1072 {LC-ESIMS:  $m/z$  1095 ([M+Na]<sup>+</sup>)}, as calculated for  $\text{C}_{55}\text{H}_{92}\text{O}_{22}$  and  $\text{C}_{54}\text{H}_{88}\text{O}_{21}$ , respectively.

In their IR spectra, the observed absorbances were consistent with the presence of olefinic double bonds. The  $^1\text{H}$  and  $^{13}\text{C}$  NMR data of compounds **5** and **6** suggested that they had similar structures, possessing the same sugar moieties but differing in their aglycones.

In the  $^1\text{H}$  NMR spectrum of compound **5**, characteristic resonances for anomeric protons were observed at  $\delta_{\text{H}}$  4.91 ( $d$ ,  $J=6.6$  Hz), 5.21 ( $d$ ,  $J=7.0$  Hz), 5.54 ( $d$ ,  $J=7.0$  Hz), 5.78 ( $bs$ ), and, in the  $^{13}\text{C}$  NMR spectrum, anomeric carbons at  $\delta_{\text{C}}$  104.2 ( $\beta$ -D-fucopyranose), 105.1 ( $\beta$ -D-glucopyranose-inner), 104.0 ( $\beta$ -D-glucopyranose-terminal) and 102.8 ( $\alpha$ -L-rhamnopyranose), as well as 2 proton signals at  $\delta_{\text{H}}$  1.35 ( $d$ ,  $J=4.8$  Hz) and 1.68 ( $d$ ,  $J=5.5$  Hz), arising from the secondary methyl groups in the sugar moieties. By means of HMBC correlations, the sequence of the saccharidic chain was determined as [ $\alpha$ -L-rhamnopyranosyl-(1 $\rightarrow$ 4)]- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 3)]- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 2)]- $\beta$ -D fucopyranoside.

The  $^1\text{H}$  NMR of compound **5** showed 6 tertiary methyl signals at  $\delta_{\text{H}}$  0.88, 0.95, 0.96, 1.07, 1.08 and 1.30. The proton signal at  $\delta_{\text{H}}$  3.21 (3H) was attributed to methoxy protons, and  $\delta_{\text{H}}$  5.46 ( $br s$ ) to the olefinic proton of the aglycone. It was determined that the aglycone was an oleanane- $\Delta^{12}$  type confirmed by presence of  $\delta_{\text{C}}$  122.6 and 148.1 signals in the  $^{13}\text{C}$  NMR spectrum. The assignment of the remaining NMR signals was achieved by means of  $^1\text{H}$ - $^1\text{H}$  COSY, HMQC, and HMBC experiments.

The location of the methoxy group was determined using HMBC correlations between methoxy protons and C-11, whereas a chemical shift of C-11 ( $\delta_{\text{C}}$  76.2) was also evident. From the chemical shift of C-11 ( $\delta_{\text{C}}$  76.2) in compound **5**, it can be concluded that the methoxyl group had an  $\alpha$ -configuration as reported for saikosaponin-b<sub>4</sub>.<sup>24</sup> The H-3 methine proton, H-23 and H-28 methylene protons showed downfield shifts due to hydroxy substitutions.

Consequently, the structure was elucidated to be 3-*O*-{[ $\alpha$ -L-rhamnosyl-(1 $\rightarrow$ 4)]- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 3)]- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 2)]- $\beta$ -D-fucopyranosyl-11-methoxy-olean-12-ene-3 $\beta$ ,23,28-triol (=ilwensisaponin C).<sup>25</sup>

Compound **6** was distinguished from compound **5** by differences in the aglycone parts in  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra.

The  $^1\text{H}$  NMR of compound **6** showed 6 tertiary methyl signals at  $\delta_{\text{H}}$  0.82, 0.87, 0.96, 0.98, 1.04 and 1.31. The olefinic protons H-11 and H-12 were determined at 5.94 ( $br d$ ,  $J=10.4$  Hz),  $\delta_{\text{C}}$  132.9 and  $\delta_{\text{H}}$  5.53 (\*),  $\delta_{\text{C}}$  131.9, respectively. Thus, aglycone was identified as an oleanane- $\Delta^{11}$  type and no signals of a methoxy group in  $^1\text{H}$  and  $^{13}\text{C}$  NMR spectra of compound **6** were observed compared with those of compound **5**.

Due to presence of an oxo-bridge between C-28 and C-13, a chemical shift of C-28 methylene protons ( $\delta_{\text{H}}$  3.33-3.72) appeared in the higher field in comparison with those of C-23 hydroxylated methylene protons ( $\delta_{\text{H}}$  3.70-4.34). Based on this evidence, the aglycone of compound **6** was determined as 13 $\beta$ ,28-epoxyolean-11-ene-3 $\beta$ ,23-diol.<sup>26</sup>

As a result, the structure of compound **6** was determined as 3-*O*-{[ $\alpha$ -L-rhamnosyl-(1 $\rightarrow$ 4)]- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 3)]- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 2)]- $\beta$ -D-fucopyranosyl}-13 $\beta$ ,28-epoxyolean-11-ene-3 $\beta$ ,23-diol (=ilwensisaponin A<sup>25</sup>=mimengoside A).<sup>27</sup>

Compound **7** (Figure 1) was obtained as an amorphous powder. Its structure was identified as verbascoside by comparing its  $^1\text{H}$  and DEPT- $^{13}\text{C}$  NMR data with previously published data and by direct comparison with the authentic sample on a TLC plate.

It has been reported that *Verbascum* L. species contained diverse iridoid glycosides such as ajugol<sup>5,13</sup>, aucubin<sup>28</sup>, lasianthoside I<sup>22</sup> and catalpol<sup>23</sup>; saponins such as ilwensisaponin C<sup>13</sup> and ilwensisaponin A<sup>13</sup>; and phenylethanoid glycosides such as verbascoside.<sup>13</sup> Ilwensisaponin A has previously been found to be active against *Aspergillus fumigatus*,<sup>29</sup> it showed moderate antifungal activity in the current study.

## CONCLUSIONS

This paper is the first to report the presence of these compounds from *V. mucronatum* Lam. Our continuing studies will be of assistance in clarifying the chemotaxonomic classification of the genus *Verbascum* L. On the other hand, when the antimicrobial activity results were evaluated, the higher activities of ajugol and ilwensisaponin A than the *V. mucronatum* extract suggest that more active compounds may be found in further phytochemical studies.

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