



# Evaluating the impact of blowing snow sea salt aerosol on springtime BrO and O<sub>3</sub> in the Arctic

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**Abstract.** We use the GEOS-Chem chemical transport model to examine the influence of bromine release from blowing snow sea salt aerosol (SSA) on springtime bromine activation and O<sub>3</sub> depletion events (ODEs) in the Arctic lower troposphere. We evaluate our simulation against observations of tropospheric BrO vertical column densities (VCD<sub>tr<sub>o</sub>p<sub>o</sub></sub>) from the GOME-2 and OMI spaceborne instruments for three years (2007-2009), as well as against surface observations of O<sub>3</sub>. We conduct a simulation with blowing snow SSA emissions from first-year sea ice (FYI, with a surface snow salinity of 0.1 psu) and multi-year sea ice (MYI, with a surface snow salinity of 0.05 psu), assuming a factor of 5 bromide enrichment of surface snow relative to seawater. This simulation captures the magnitude of observed March-April GOME-2 and OMI VCD<sub>tr<sub>o</sub>p<sub>o</sub></sub> to within 17%, as well as their spatiotemporal variability ( $r=0.76-0.85$ ). Many of the large-scale bromine explosions are successfully reproduced, with the exception of events in May, which are absent or systematically underpredicted in the model. If we assume a lower salinity on MYI (0.01 psu) some of the bromine explosions events observed over MYI are not captured, suggesting that blowing snow over MYI is an important source of bromine activation. We find that the modeled atmospheric deposition onto snow-covered sea ice becomes highly enriched in bromide, increasing from enrichment factors of  $\sim 5$  in September-February to 10-60 in May, consistent with freshly fallen snow composition observations. We propose that this progressive enrichment in deposition could enable blowing snow-induced halogen activation to propagate into May and might explain our late-spring underestimate in VCD<sub>tr<sub>o</sub>p<sub>o</sub></sub>. We estimate that atmospheric deposition of SSA could increase snow salinity by up to 0.04 psu between February and April, which could be an important source of salinity for surface snow on MYI as well as FYI covered by deep snowpack. Inclusion of halogen release from blowing snow SSA in our simulations decreases monthly mean Arctic surface O<sub>3</sub> by 4-8 ppbv (15-30%) in March and 8-14 ppbv (30-40%) in April. We reproduce a transport event of depleted O<sub>3</sub> Arctic air down to 40° N observed at many sub-Arctic surface sites in early April 2007. While our simulation captures a few ODEs observed at coastal Arctic surface sites, it underestimates the magnitude of other events and entirely misses some events. We suggest that inclusion of direct snowpack activation, which is a strong local source of Br radicals in the shallow

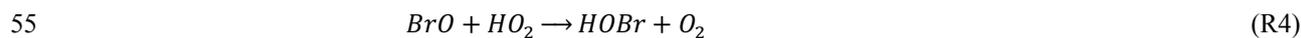


35 Arctic boundary layer, could help reconcile the success of our simulation at capturing satellite retrievals of  $VCD_{\text{tropo}}$  with its difficulty in reproducing local ODEs.

## 1 Introduction

Ozone depletion events (ODEs) are often observed in the springtime Arctic boundary layer (Barrie et al., 1988; Bottenheim et al., 2009, 1986; Bottenheim and Chan, 2006; Halfacre et al., 2014; Koo et al., 2012; Oltmans et al., 2012; Oltmans and Komhyr, 40 1986). While ODEs occur episodically at coastal Arctic sites, lasting 1-3 days, they are more persistent and widespread over the frozen ocean (e.g. Simpson et al., 2007 and references therein). These ODEs have been linked to the release of significant levels of bromine radicals, in a phenomenon called bromine explosion (Abbatt et al., 2012; Simpson et al., 2007b). Both ground-based and satellite instruments have reported elevated BrO columns over the Arctic during March and April (Choi et al., 2012, 2018; Hönninger and Platt, 2002; Jones et al., 2009; Liao et al., 2012; Neuman et al., 2010; Peterson et al., 2017; 45 Salawitch et al., 2010; Simpson et al., 2017; Theys et al., 2011; Wagner et al., 2001). In addition to ODEs in surface air during spring, the elevated bromine radicals result in the rapid oxidation of mercury and its deposition to snowpack (Ebinghaus et al., 2002; Schroeder et al., 1998; Steffen et al., 2008), which can have significant impacts on the health of people and wildlife in polar regions (AMAP, 2011).

An autocatalytic cycle involving the heterogeneous release of sea salt bromide ( $\text{Br}^-$ ) via uptake of HOBr was suggested to be 50 the primary cause of bromine explosions and ODEs (Abbatt et al., 2012; Fan and Jacob, 1992; Hausmann and Platt, 1994; Simpson et al., 2007b):



Reaction (R1) is a multiphase reaction, which takes place on acidic aerosols, cloud droplets, and snow grains. This reaction is required for the initial release of bromine to the atmosphere, but also maintains high levels of BrO by allowing fast recycling 60 of  $\text{Br}^-$  to reactive bromine (Fan and Jacob, 1992; Lehrer et al., 2004). In addition, the heterogeneous reaction of HOBr with sea salt chloride ( $\text{Cl}^-$ ) can release BrCl from the condensed phase (Abbatt et al., 2012).



In the global troposphere, inorganic bromine ( $\text{Br}_y$ ) has three major sources: debromination of sea salt aerosol (SSA) produced by breaking waves in the open ocean, photolysis and oxidation of bromocarbons, and transport of  $\text{Br}_y$  from the stratosphere. Release of  $\text{Br}^-$  from oceanic SSA is estimated to be the largest global source of tropospheric  $\text{Br}_y$  (Sander et al., 2003; Zhu et al., 2019). Current global models that include these three sources of tropospheric bromine as well as multiphase reactions such as (R1), cannot reproduce the observed elevated levels of tropospheric  $\text{BrO}$  over polar regions during spring (Parrella et al., 2012; Schmidt et al., 2016). Three classes of substrates specific to polar regions have been proposed as a source of  $\text{Br}^-$  in reaction (R1): salty snowpack on sea ice and coastal regions (McConnell et al., 1992; Simpson et al., 2005; Toyota et al., 2011), first-year sea ice (Frieß et al., 2004; Nghiem et al., 2012; Wagner et al., 2007), SSA produced from frost flowers (Kaleschke et al., 2004; Rankin et al., 2002), and SSA produced from blowing salty snow (Jones et al., 2009; Yang et al., 2008).

Both laboratory and outdoor chamber experiments have detected  $\text{Br}_2$  production when acidified surface saline snow was exposed to sunlight (Pratt et al., 2013; Wren et al., 2013). However, no  $\text{Br}_2$  production was detected over sea ice and brine icicles, suggesting that brines and frost flowers on new sea ice surfaces are not a bromine activation source (Pratt et al., 2013). The proposed role of frost flowers as a direct source of SSA is disputed by several field and laboratory experiments showing that frost flowers are difficult to break even under strong wind conditions (e.g., Alvarez-Aviles et al., 2008; Roscoe et al., 2011; Yang et al., 2017).

A number of studies have focused on the role of aerosols, in particular SSA from blowing snow, as a key mechanism to initiate and sustain bromine activation aloft. Yang et al. (2008) proposed blowing snow SSA, produced after sublimation of wind-lofted salty snow on sea ice, as a source of  $\text{BrO}$ . This source is consistent with satellite observations of large-scale  $\text{BrO}$  column enhancements over Arctic and Antarctic sea ice, which are often associated with high surface wind speeds that can generate blowing snow events (Begoïn et al., 2010; Blechschmidt et al., 2016; Choi et al., 2018; Jones et al., 2009). Using ground-based measurements in coastal Alaska, Frieß et al. (2004) found that periods of enhanced  $\text{BrO}$  were coincident with an increase in aerosol extinction at higher wind speeds ( $> 5 \text{ m s}^{-1}$ ), suggesting that halogen activation and/or recycling takes place in situ on SSA aerosol produced by sublimation of dispersed snow grains or frost flowers. Peterson et al. (2017) and Simpson et al. (2017) reported aircraft observations of the vertical distribution of  $\text{BrO}$  and aerosol extinction consistent with initial activation on snowpack followed by transport aloft (500-1000 m), where high  $\text{BrO}$  was sustained by recycling on aerosols. More recently, Frey et al. (2019) and Giordano et al. (2018) reported direct observations of SSA production from blowing snow above sea ice. In particular, Frey et al. (2019) found that the  $\text{Br}^-/\text{Na}^+$  ratio of blowing snow SSA observed at 29 m above the ground decreased by a factor of 2-3 relative to observations at 2 m, suggesting rapid  $\text{Br}^-$  release via (R1) from blowing snow SSA.

There are only a few modeling studies which have examined the link between blowing snow SSA and bromine explosions. In their pioneering study, Yang et al. (2010) implemented the Yang et al. (2008) blowing snow parameterization in the p-TOMCAT model and presented a comparison to satellite retrievals of  $\text{BrO}$  columns from the Global Ozone Monitoring



Experiment (GOME) for 2 months. Their evaluation showed qualitative agreement with observations over polar regions for  
95 those months. Theys et al. (2011) describe two 3-day case studies over the Arctic and Antarctic comparing the p-TOMCAT  
simulation to tropospheric BrO columns from the second Global Ozone Monitoring Experiment (GOME-2). There was  
qualitative agreement with the location and timing of the two BrO explosion events. However, Theys et al. (2011) note that  
beyond these case studies there were large discrepancies between model and observations, reflecting uncertainties in the p-  
TOMCAT Br<sup>-</sup> emissions, such as assumptions about snow salinity and fraction of Br<sup>-</sup> in SSA released to the gas phase. Zhao  
100 et al. (2016) used the UM-UKCA chemistry-climate model to simulate a bromine explosion case study initiated by a blowing  
snow event and transported to the Canadian high Arctic. While the model reproduced the vertical extent of the BrO plume, it  
was not as successful at capturing the timing or magnitude of the observed BrO. Choi et al. (2018) demonstrated a strong  
spatial and temporal correlation between bromine explosions observed by OMI and blowing snow SSA as simulated in the  
GEOS-5 modeling system. However, their study did not include a simulation of bromine photochemistry.

105 Here, we use the GEOS-Chem global chemical transport model to further quantify the role played by blowing snow SSA in  
springtime Arctic bromine explosions and examine the associated tropospheric O<sub>3</sub> depletion. We systematically evaluate a 3-  
year GEOS-Chem simulation (2007-2009) via comparisons to satellite retrievals of tropospheric BrO vertical column densities  
(VCD<sub>tropo</sub>) from GOME-2 and the Ozone Monitoring Instrument (OMI). We also compare the model to in situ surface  
observations of O<sub>3</sub> at several Arctic and sub-Arctic sites. Section 2 describes the GEOS-Chem simulations and the observations  
110 used in this study. We evaluate the model's capability to reproduce the timing, magnitude, and spatial extent of bromine  
explosion events observed by GOME-2 and OMI in Section 3. In section 4, we examine the impact of blowing snow SSA on  
surface O<sub>3</sub>. We assess the contribution of atmospheric deposition to surface snow salinity and bromide content in Section 5.  
Conclusions are presented in Section 6.

## 2 Observations and model simulations

### 115 2.1 Satellite observations of tropospheric BrO vertical column densities

The Second Global Ozone Monitoring Experiment (GOME-2) is a nadir-scanning UV/visible spectrometer on the METOP-A  
satellite, which was launched on 19 October 2006 in a sun-synchronous polar orbit with an equator crossing time of 9:30 LT  
(Munro et al., 2006). The GOME-2 spectrometer covers the 240-790 nm wavelength region, with a spectral resolution between  
0.26 and 0.51 nm. It has a ground pixel size of 80 km × 40 km and a scanning swath of 1920 km. In this study, we use the  
120 daily 2007-2009 GOME-2 VCD<sub>tropo</sub> retrieved by Theys et al. (2011). Briefly, the VCD<sub>tropo</sub> are derived from a residual technique  
that combines measured slant columns and calculated stratospheric columns, accounting for the impact of clouds, surface  
reflectivity and viewing geometry on the measurement sensitivity. The stratospheric BrO contribution is removed using the  
dynamic climatology based on the BASCOE CTM as described in Theys et al. (2009, 2011). The tropospheric slant columns  
are converted to vertical columns with a tropospheric air mass factor (AMF) assuming different BrO profile shapes depending  
125 on surface albedo. For low surface albedo (<0.5), a Gaussian shape BrO profile with a maximum at 6 km is used. For high



130 surface albedo ( $>0.5$ ), the assumed tropospheric BrO concentration profile is constant in the first km above the Lambertian surface reflector. For this second case, which characterizes most of the ice- and snow-covered Arctic, there is a high sensitivity to BrO close to the surface and a weak dependence of the tropospheric AMF to the shape of the profile. As in Theys et al. (2011), we only consider retrievals for solar zenith angles (SZA) less than  $80^\circ$  and cloud fractions below 0.4 for which the pressure difference between the surface and the top of the cloud less than 400 mbar.

We also use the BrO  $VCD_{\text{tropo}}$  from the Ozone Monitoring Instrument (OMI) onboard the Aura satellite, which was launched on 15 July 2004 in a sun-synchronous polar orbit with a 13:30 LT equator overpass time (Levelt et al., 2006). OMI is a nadir solar backscatter spectrometer that measures ultraviolet–visible wavelengths (270–500 nm), and has a horizontal resolution of  $13 \text{ km} \times 14 \text{ km}$  and swath width of 2600 km. Since 2008, the OMI swath coverage has been reduced due to an external  
135 obstruction. We use here the daily 2007–2009  $VCD_{\text{tropo}}$ , which were retrieved by Choi et al. (2018) for the months of March and April poleward of  $60^\circ\text{N}$ . The total BrO slant column densities are derived from the OMI total BrO (OMBRO) product, by fitting a model function to OMI UV backscattered radiance at 319–347.5 nm (Choi et al., 2018). The  $VCD_{\text{tropo}}$  are retrieved using the same residual technique and dynamic climatology stratospheric BrO columns as in Theys et al. (2011). In the OMI tropospheric AMF calculation, the assumed BrO shape profile is based on a composite of aircraft measurements obtained  
140 during the NASA Arctic Research of the Composition of the Troposphere from Aircraft and Satellites (ARCTAS) campaign (Choi et al., 2012). We use the same selection criteria as in Choi et al. (2018):  $\text{SZA} < 80^\circ$ , surface reflectivity  $> 0.6$ , and only retain observations with low cloud contamination (difference between OMI rotational Raman cloud pressure and terrain pressure  $< 100 \text{ hPa}$ ). The uncertainty of OMI and GOME-2  $VCD_{\text{tropo}}$  retrieved over the highly reflective surfaces of polar regions is on the order of 30–50%.

## 145 2.2 Surface $\text{O}_3$ measurements

We use hourly in situ  $\text{O}_3$  measurements at three Arctic surface sites: Utqiagvik, Alaska (also known as Barrow,  $71.3^\circ \text{N}$ ,  $156.6^\circ \text{W}$ , 8 m above sea level) from the NOAA Earth System Research Laboratory, Alert, Nunavut ( $82.5^\circ \text{N}$ ,  $62.5^\circ \text{W}$ , 187 m) from the Canadian Air and Precipitation Monitoring Network (CAPMoN), and Zeppelin, Spitzbergen ( $78.9^\circ \text{N}$ ,  $11.8^\circ \text{E}$ , 474 m) from the Norwegian Institute for Air Research. We also use hourly surface  $\text{O}_3$  measurements at several sub-arctic sites from  
150 CAPMoN as well as from the Canadian National Air Pollution Surveillance (NAPS) network and from the United States Clean Air Status and Trends Network (CASTNET). For the NAPS sites, we only consider sites sampling background air in categories “Forest” and “Undeveloped Rural”, and for CASTNET we exclude sites located in “Urban/Agricultural” areas. This selection allows us to avoid more polluted sites, which can be subject to low winter-spring  $\text{O}_3$  concentrations associated with NO titration of  $\text{O}_3$ . The subarctic sites include Bonner Lake ( $49.4^\circ \text{N}$ ,  $82.1^\circ \text{W}$ , 242 m), Algoma ( $47.0^\circ \text{N}$ ,  $84.38^\circ \text{W}$ , 411 m), and Egbert  
155 ( $44.2^\circ \text{N}$ ,  $79.8^\circ \text{W}$ , 206 m) from the CAPMoN network, as well as Elk Island ( $53.68^\circ \text{N}$ ,  $112.87^\circ \text{W}$ , 714 m) from NAPS and Woodstock ( $43.94^\circ \text{N}$ ,  $71.70^\circ \text{W}$ , 255 m) from CASTNET.



### 2.3 The GEOS-Chem chemical transport model

160 GEOS-Chem is a 3-D global chemical transport model (Bey et al., 2001). In this work, we use GEOS-Chem v11-02d  
(<http://www.geos-chem.org>, last access 19 August 2019), driven by the Modern-Era Retrospective analysis for Research and  
Applications, version 2 (MERRA-2) assimilated meteorological fields (Gelaro et al., 2017), which have a native horizontal  
resolution of 0.5° latitude by 0.625° longitude with 72 vertical levels. We regrid the MERRA-2 fields to a 2°× 2.5° horizontal  
resolution and 47 vertical levels with merged levels above 80 hPa for computational expediency. For the time-period of the  
simulations conducted here (2007-2009), the daily boundary conditions for sea ice concentrations in MERRA-2 are from the  
high-resolution (1/20°) Operational SST and Sea Ice Analysis (OSTIA), which uses daily sea ice concentration products from  
165 multiple Special Sensor Microwave Imager (SSM/I) satellites (Donlon et al., 2012).

Global anthropogenic emissions are from EDGAR v4.2 (Olivier and Berdowski, 2001) for 1970-2008. Biomass burning  
emissions are from the Global Fire Emissions Database version 4 (GFEDv4) emission inventory (van der Werf et al., 2017).  
Biogenic emissions of volatile organic compounds (VOCs) are from the Model of Emissions of Gases and Aerosols from  
Nature version 2.1 (MEGAN 2.1) (Guenther et al., 2012).

170 GEOS-Chem simulates detailed HO<sub>x</sub>-NO<sub>x</sub>-VOC-O<sub>3</sub>-halogen-aerosol tropospheric chemistry. The chemical mechanism in  
GEOS-Chem v11-2d was updated to the most recent JPL/IUPAC recommendations, as described in (Fischer et al., 2014;  
Fisher et al., 2016; Mao et al., 2013; Travis et al., 2016). This version of GEOS-Chem includes bromine-chlorine-iodine  
halogen chemistry. The bromine chemistry mechanism was first described in Parrella et al. (2012). The mechanism was then  
updated by Schmidt et al. (2016) to include extensive multiphase chemistry as well as coupling to tropospheric chlorine  
175 chemistry, which provides an important pathway to recycle bromine radicals. The uptake coefficients for the heterogeneous  
reactions of HOBr, ClNO<sub>3</sub> and O<sub>3</sub> with Br<sup>-</sup> in aerosols, as well as for the reaction of HOBr with Cl<sup>-</sup> in aerosols follow Ammann  
et al. (2013). The uptake coefficient between O<sub>3</sub> and Br<sup>-</sup> considers both bulk and surface reactions. Sherwen et al. (2016a,  
2016b) implemented iodine chemistry and Cl-Br-I interactions. Chen et al. (2017) added the in-cloud oxidation of dissolved  
SO<sub>2</sub> (S(IV)) by HOBr in GEOS-Chem, which decreased the global Br<sub>y</sub> burden by 50% and resulted in improved agreement  
180 with GOME-2 VCD<sub>tropo</sub> between 60° N and 60° S. Fifteen bromine tracers are transported (Br<sub>2</sub>, Br, BrO, HOBr, HBr, BrNO<sub>2</sub>,  
BrCl, BrONO<sub>2</sub>, CHBr<sub>3</sub>, CH<sub>2</sub>Br<sub>2</sub>, CH<sub>3</sub>Br, IBr, CH<sub>2</sub>IBr, Br<sup>-</sup> on accumulation mode SSA and Br<sup>-</sup> on coarse mode SSA), with  
sources from photolysis of CHBr<sub>3</sub>, oxidation of CHBr<sub>3</sub>, CH<sub>2</sub>Br<sub>2</sub> and CH<sub>3</sub>Br by OH radicals, transport of reactive bromine from  
the stratosphere, and SSA debromination driven by explicit heterogeneous reactions of SSA Br<sup>-</sup> with HOBr, ClNO<sub>3</sub>, and O<sub>3</sub>.

Open ocean emissions of SSA are a function of wind speed and sea surface temperature (SST) as described in Jaeglé et  
185 al.(2011), with updates from Huang and Jaeglé (2017) for cold ocean waters (SST < 5°C). Two separate SSA tracers are  
transported: accumulation mode SSA ( $\tau_{dry} = 0.01-0.5 \mu\text{m}$ ) and coarse mode SSA ( $\tau_{dry} = 0.5-8 \mu\text{m}$ ). Sea salt Br<sup>-</sup> is emitted



assuming a ratio of  $2.11 \times 10^{-3}$  kg Br per kg of dry SSA, based on the composition of sea water (Lewis and Schwartz, 2004; Sander et al., 2003). Sea salt  $\text{Br}^-$  is transported in two tracers as part of accumulation and coarse mode SSA.

### 2.3.1 The blowing snow SSA simulation in GEOS-Chem

190 The blowing snow SSA simulation in GEOS-Chem is described in Huang and Jaeglé (2017) and Huang et al. (2018). Blowing  
snow SSA emissions are a function of relative humidity, temperature, age of snow, surface snow salinity and wind speed  
following the parameterization of Yang et al. (2008, 2010). As in Huang and Jaeglé (2017), we assume  $N=5$  for the number of  
SSA particles produced per snowflake, and a mean snow age of 3 days over the Arctic. Two key parameters controlling the  
magnitude of blowing snow SSA production and subsequent heterogeneous  $\text{Br}^-$  release are the surface snow salinity and  $\text{Br}^-$   
195 content, both of which have very few observational constraints. Sea salt and  $\text{Br}^-$  in surface snow originates from upward  
migration of brine from sea ice, atmospheric deposition of SSA as well as gas-phase reactive bromine species, and  
contamination of snow by frost flowers (Abbatt et al., 2012). Domine et al. (2004) estimates that frost flowers could account  
for 10% of the observed surface snow salinity. Upward migration of brine is expected to be the dominant source of salinity for  
thin snow over sea ice ( $<10$ - $17$  cm) (Domine et al., 2004; Peterson et al., 2019). As the snow depth gets thicker, the salinity of  
200 surface snow decreases and the influence of atmospheric deposition likely becomes more important (Krnavek et al., 2012;  
Nandan et al., 2017). Snow over first-year sea ice (FYI) is typically more saline than over multi-year sea ice (MYI), as MYI  
is desalinated by flushing and gravity drainage during repeated summer melt cycles.

Krnavek et al. (2012) sampled the chemical composition of surface snow on land-fast sea ice near Utqiagvik, Alaska. They  
reported a median salinity of 0.7 practical salinity unit (psu) for 2-3 weeks old FYI, 0.1 psu for thicker FYI, and 0.01 psu for  
205 MYI. Domine et al. (2004) reported a salinity of 0.02 psu for MYI near Alert. Peterson et al. (2019) measured surface snow in  
FYI and MYI regions near Greenland, Alaska, as well as in the central Arctic, finding higher mean salinities for snowpacks  
less than 17 cm deep (0.15 psu) compared to deeper snowpacks (0.02 psu). The observed  $\text{Br}^-$  concentrations in surface snow  
over Arctic sea ice is highly variable, ranging from  $10^{-2}$  to  $10^3$   $\mu\text{M}$  (Domine et al., 2004; Krnavek et al., 2012; Peterson et al.,  
2019). This high variability is accompanied with either depletion or enhancement in the bromide to sodium ( $\text{Br}^-/\text{Na}^+$ ) ratio  
210 relative to sea water composition. Depletion in  $\text{Br}^-$  relative to seawater indicates loss to the atmosphere via heterogeneous  
reactions. Enhancements can be the result of precipitation of hydrohalite ( $\text{NaCl} \cdot 2\text{H}_2\text{O}$ ) below 251K in brine (Koop et al., 2000;  
Morin et al., 2008). Indeed, the laboratory measurements of aqueous NaCl and sea-salt solution droplets of Koop et al. (2000)  
show a factor of 12 increase in the  $\text{Br}^-/\text{Na}^+$  ratio at 240K relative to 273K. Enhancements in  $\text{Br}^-$  could also be caused by the  
deposition of atmospheric aerosol and HBr produced during bromine explosions onto surface snow (Simpson et al., 2005,  
215 2007b). Domine et al. (2004) report measurements of surface snow on sea ice near Alert, with a factor of 5 enrichment in  $\text{Br}^-$   
relative to seawater, while surface snow over an Arctic Ocean site further north displayed a  $\text{Br}^-$  enrichment of 25. Samples of  
surface snow on MYI appear to be enhanced in  $\text{Br}^-$  more often than on FYI (Krnavek et al., 2012), which suggests that MYI,  
in addition to FYI, could play an active role in Arctic boundary layer bromine and chlorine chemistry (Peterson et al., 2019).



We conduct three GEOS-Chem simulations as part of this work: a standard simulation (referred to as “STD”) in which the  
220 only source of SSA is from the open ocean, and two simulations in which we add a blowing snow source of SSA. In the first  
blowing snow simulation (referred to as “FYI Snow”), we assume an Arctic surface snow salinity of 0.1 psu on FYI and 0.01  
psu for MYI. In the second blowing snow simulation (“FYI+MYI Snow”), we use a salinity of 0.1 psu on FYI and 0.05 psu  
on MYI. These two salinities assumed for MYI are used to examine the role of bromine activation on MYI. For both blowing  
snow simulations, we assume a surface snow  $\text{Br}^-$  enrichment factor of 5 relative to sea water (sea salt  $\text{Br}^-$  is emitted as assuming  
225 a ratio of  $10.55 \times 10^{-3}$  kg Br per kg of dry blowing snow SSA emitted). As sea ice age is not tracked in MERRA-2, for each  
year we identify the location of Arctic MYI from the preceding September minimum sea ice extent in MERRA-2. The FYI  
extent is calculated by subtracting the MYI extent from the total sea ice extent. Note that this very simple approach does not  
account for advection and melting of MYI between one September to the next. All simulations are initialized with a 6-month  
spin-up in 2006 and then followed by a 3-year simulation for 2007-2009.

230 In previous work (Huang et al., 2018; Huang and Jaeglé, 2017), we found that the GEOS-Chem blowing snow simulation  
(using 0.1 psu on FYI and 0.01-0.1 psu on MYI) reproduced the seasonal cycle and magnitude of SSA mass concentrations  
observed at Utqiagvik, Alert, and Zeppelin, but that the STD simulation underestimates observations by a factor of 2-10 during  
winter and spring. The blowing snow simulations also reproduced the seasonal cycle of aerosol extinction coefficients  
observed in the lower troposphere (0-2 km) over Arctic sea ice by the Cloud-Aerosol Lidar with Orthogonal Polarization  
235 (CALIOP) on-board the CALIPSO satellite (Huang et al., 2018).

For comparison with satellite retrievals of  $\text{VCD}_{\text{tropo}}$ , we sample the model at 9:00-11:00 LT, which corresponds to the overpass  
times of GOME-2 in the Northern Hemisphere. We found that sampling GEOS-Chem at 13:00-15:00 LT, matching the OMI  
overpass time, results in less than a 1% difference in  $\text{VCD}_{\text{tropo}}$ . We thus only show the model results for 9:00-11:00 LT. We  
regrid daily OMI and GOME-2  $\text{VCD}_{\text{tropo}}$  to a horizontal resolution of  $2^\circ \times 2.5^\circ$  for comparison to GEOS-Chem. The model is  
240 sampled on days and locations of GOME-2 and OMI overpasses with available retrievals.

### 3 Evaluation of the impact of blowing snow SSA on BrO $\text{VCD}_{\text{tropo}}$

#### 3.1 Seasonal cycle of $\text{VCD}_{\text{tropo}}$ in the Northern Hemisphere

Several previous GEOS-Chem model versions have been evaluated against GOME-2 BrO  $\text{VCD}_{\text{tropo}}$ . Schmidt et al. (2016)  
showed that the inclusion of SSA debromination resulted in a 50-100% overestimate in BrO in the northern hemisphere.  
245 Sherwen et al. (2016a, 2016b) added iodine chemistry and disabled SSA debromination, finding relatively good agreement  
with GOME-2 observations over the Arctic and the summertime low- and mid- latitudes. Chen et al. (2017) enabled SSA  
debromination and added in-cloud oxidation of dissolved S(IV) with HOBr to the model version of Schmidt et al. (2016)  
without iodine chemistry. They found improved agreement with GOME-2  $\text{VCD}_{\text{tropo}}$  over low- and mid- latitudes, but factors



of 3-10 underestimate over high latitudes. Our STD simulation includes SSA debromination, in-cloud oxidation of dissolved  
250 S(IV) by HOBr, as well as iodine chemistry, and chlorine chemistry.

Fig. 1 (panels a and b) shows that with these four components, the STD simulation agrees well with the GOME-2  $VCD_{\text{tropo}}$  at  
0-30° N and 30-60° N averaged for 2007-2009. Over the Arctic (>60° N), however, the STD simulation underestimates  
GOME-2 and OMI observations by up to 50% during spring (Fig. 1c). Note that we do not show monthly mean GOME-2  
255  $VCD_{\text{tropo}}$  at 60-90° N for November-February, as less than 70% of the polar region has valid data (SZA is generally greater  
80° for these months).

The 2007-2009 GOME-2 and OMI  $VCD_{\text{tropo}}$  display a March-April maximum of  $\sim 3\text{-}3.5 \times 10^{13} \text{ cm}^{-2}$  poleward of 60° N (Fig.  
1c). The inclusion of blowing snow SSA emissions in the GEOS-Chem FYI Snow simulation increases the modeled springtime  
 $VCD_{\text{tropo}}$  by 42% in March and 52% in April, improving the agreement with satellite retrievals (Fig. 1c). The FYI+MYI Snow  
simulation increases the modeled  $VCD_{\text{tropo}}$  by another 10-20% in March and April. Both blowing snow simulations are within  
260 5-20% of observed  $VCD_{\text{tropo}}$  in March and April, however they predict too rapid a decrease in  $VCD_{\text{tropo}}$  in May.

In late summer and fall, when sea ice extent is at its minimum and blowing snow SSA emissions are negligible, GEOS-Chem  
underestimates GOME-2  $VCD_{\text{tropo}}$  by 30-40%. Over the cloudy summer Arctic, the reaction  $\text{HOBr} + \text{S(IV)}$  in cloud water  
provides a sink for HOBr and decreases the  $\text{Br}_y$  abundance by about 70–90% (Chen et al., 2017). We hypothesize that including  
acid displacement HCl from SSA would lead to an increase in HOBr because of the competition between  $\text{HOBr} + \text{Cl}^-$  (which  
265 recycles HOBr by producing BrCl) and  $\text{HOBr} + \text{S(IV)}$  (which is a sink for HOBr) in cloud droplets. Indeed, in a subsequent  
version of GEOS-Chem, Wang et al. (2018) added this source of HCl, finding an increase in BrO, especially in cloudy high  
latitudes.

### 3.1 Spatial distribution of springtime Arctic BrO $VCD_{\text{tropo}}$

Fig. 2 shows the spatial distribution of monthly mean  $VCD_{\text{tropo}}$  during March and April 2007-2009. In March, both GOME-2  
270 and OMI exhibit enhanced  $VCD_{\text{tropo}}$  ( $3.5\text{-}5 \times 10^{13} \text{ cm}^{-2}$ ) in an arc between Baffin Bay - along the west coast of Greenland - and  
the Laptev Sea, off the Northern coast of Siberia. In April, the observed  $VCD_{\text{tropo}}$  reach values  $> 3.5\text{-}4 \times 10^{13} \text{ cm}^{-2}$  over most of  
the sea ice covered Arctic Ocean. Fig. 3a compares the GOME-2 and OMI  $VCD_{\text{tropo}}$  poleward of 60°N for March and April  
2007-2009, showing that they are highly correlated ( $r=0.92$ , slope=0.99), with OMI being 13% higher than GOME-2.

As shown in Figs. 2 and 3d, the GEOS-Chem STD simulation predicts  $VCD_{\text{tropo}}$  lower than  $2.5\text{-}3 \times 10^{13} \text{ cm}^{-2}$  in March-April,  
275 with little spatial variability compared to observations ( $r=0.22\text{-}0.36$ ). The normalized mean bias ( $\text{NMB} = 100 \times \sum(M_i - O_i) / \sum O_i$ ,  
with observations  $O_i$  and model  $M_i$  summed poleward of 60° N) is -41% compared to OMI and -34% compared to GOME-2.  
The FYI Snow simulation has more success at reproducing the magnitude of the observed  $VCD_{\text{tropo}}$  in March and April  
( $\text{NMB}=-12\%$  relative to OMI and  $\text{NMB}=+1\%$  relative to GOME-2), as well as their spatial distribution ( $r=0.70\text{-}0.76$ ). The



VCD<sub>tropo</sub> predicted by the FYI+MYI Snow simulation agrees better with OMI (NMB=+2% and  $r=0.76$ ), but overestimates  
280 GOME-2 (NMB=+17% and  $r=0.79$ ). Similar to the satellite retrievals in March, the FYI and FYI+MYI snow simulations  
displays the largest VCD<sub>tropo</sub> over the Canadian Arctic Archipelago (Fig. 2), with secondary maxima over Baffin Bay, the  
Greenland Sea, and the east Siberian Sea. In April, the VCD<sub>tropo</sub> in the FYI snow simulation underestimates GOME-2 and OMI  
throughout the Arctic, but the FYI+MYI snow simulation overestimates satellite observations in the high Arctic ( $>80^\circ$  N),  
while underestimating observations at  $70\text{--}80^\circ$  N. This suggests spatial variability in salinity and/or Br<sup>-</sup> enhancement for snow  
285 that is not captured in our assumptions of spatially uniform values for FYI and MYI. We also find that the simulated VCD<sub>tropo</sub>  
underestimate observations over the coastal Arctic, especially in April. This could be due halogen activation from snow over  
land (Simpson et al., 2005), which is not considered in our simulation as we assume zero salinity and Br<sup>-</sup> content in continental  
snow.

### 3.2 Daily variations in pan-Arctic BrO VCD<sub>tropo</sub>

290 Fig. 4a shows that GOME-2 daily mean VCD<sub>tropo</sub> poleward of  $60^\circ$  N is highest in March-April, and then decreases in May.  
The OMI VCD<sub>tropo</sub> in March and April displays a similar day-to-day variability as GOME-2. Overall the FYI+MYI Snow  
simulation shows good agreement with the magnitude and daily variability of the combined mean GOME-2 and OMI VCD<sub>tropo</sub>  
(NMB=-8.3%,  $r=0.85$ ). While the FYI+MYI Snow simulation reproduces the March-April VCD<sub>tropo</sub>, it predicts too rapid a  
decrease in May for all three years (Fig 4a). The difference between the FYI Snow and MYI+FYI Snow simulations is the  
295 largest in April as polar sunrise reaches the high Arctic, where the majority of MYI is located. Most of the BrO enhancements  
in the two blowing snow simulations occur below 1 km altitude, occasionally extending to 2 km (see Fig. S1ab).

As discussed in previous studies (Choi et al., 2018; Nghiem et al., 2012; Richter et al., 1998; Salawitch et al., 2010; Theys et  
al., 2011) satellite observations show the frequent occurrence of “BrO hotspots” over Arctic sea ice during spring. For this  
study, we define a BrO hotspot when local VCD<sub>tropo</sub> exceed  $4.5 \times 10^{13}$  cm<sup>-2</sup>, which is the 90<sup>th</sup> percentile of GOME-2 VCD<sub>tropo</sub>  
300 poleward of  $60^\circ$  N in March-April. For OMI, the corresponding 90<sup>th</sup> percentile is  $5.1 \times 10^{13}$  cm<sup>-2</sup>. We use these respective  
criteria to calculate the daily area covered by BrO hotspots as observed by GOME-2 and OMI. The episodic nature of GOME-  
2 and OMI BrO hotspots is apparent in Fig. 4b, which shows a large degree of daily variability, with events lasting between a  
few days to 2 weeks. The green shading in Fig. 4 identifies multi-day events ( $>5$  days) when the GOME-2 hotspot area exceeds  
2 million km<sup>2</sup>. There are 2 such events in both 2007 and 2008, and 3 events in 2009 (Fig. 4b). These events reach maximum  
305 extents of 4-6 million km<sup>2</sup>. For comparison, the mean sea-ice extent during March-April poleward of  $60^\circ$ N is  $\sim 10$  million km<sup>2</sup>  
(with a  $\sim 65\%/35\%$  split between FYI and MYI).

We apply the OMI threshold (VCD<sub>tropo</sub>  $> 5.1 \times 10^{13}$  cm<sup>-2</sup>) to the GEOS-Chem simulations. On average, the FYI+MYI Snow  
simulation reproduces the observed spatial extent of BrO hotspots to within 4% and captures the timing of the 2 large-scale  
episodes in 2007 (26 March – 4 May, 10–20 May 2007), the two episodes in 2008 (7–17 March and 1–8 April 2008). For



310 2009, the FYI+MYI Snow simulation predicts two episodes (5–15 March and 11–23 April), but the aerial extent of these episodes is overestimated, and the simulation misses the third episode on 6–19 May 2009 (Fig. 4b). In the model, the variability in BrO hotspots is driven by the temporal variations of the blowing snow SSA burden (Fig. 4c).

### 3.3 Two case studies of large BrO explosion events

Fig. 5 shows the daily spatial distribution of  $VCD_{\text{tropo}}$  on 25–30 March 2007, corresponding the largest BrO hotspot episode  
315 observed by GOME-2 and OMI during our study period (Fig. 4ab). The GOME-2 and OMI  $VCD_{\text{tropo}}$  exceed  $7 \times 10^{13} \text{ cm}^{-2}$  (up to  $15 \times 10^{13} \text{ cm}^{-2}$ ) over the sea-ice covered region in the high Arctic region ( $>70^\circ\text{N}$ ). The enhanced  $VCD_{\text{tropo}}$  start on 25 March over the East Siberian Sea and then rotate counterclockwise, reaching the Beaufort Sea on 27 March and the Canadian Archipelago on 28–30 March 2007. This event was previously discussed in Begoin et al. (2010) and Choi et al. (2018). Begoin et al. (2010) used the FLEXPART particle dispersion model to link the enhanced GOME-2  $VCD_{\text{tropo}}$  to a cyclone with very  
320 high surface wind speeds, favorable for generating blowing snow SSA. Choi et al. (2018) found that the spatial pattern of OMI  $VCD_{\text{tropo}}$  was consistent with that of the GEOS-5 simulated blowing snow SSA burden on both FYI and MYI, while restricting the blowing snow emissions to FYI only did not agree as well with the observed  $VCD_{\text{tropo}}$ . Our FYI+MYI Snow simulation reproduces the evolution of this BrO hotspot, although the magnitude of the modeled  $VCD_{\text{tropo}}$  overestimates observations after 28 March 2007. The main difference between the FYI Snow and FYI+MYI Snow simulations is on 25–26 March 2007,  
325 when the high winds are located over the North Pole where MYI is (Figs. 5 and S2). As there are no valid GOME-2 and OMI retrievals at these high latitudes ( $\text{SZA} > 80^\circ$ ), we cannot assess which simulation agrees better with observations for this case study.

Fig. 6 shows the daily distribution of  $VCD_{\text{tropo}}$  on 14–19 April 2007. During this event, there is a factor of 1.5 difference in pan-Arctic  $VCD_{\text{tropo}}$  calculated with the FYI Snow and FYI+MYI Snow simulations (Fig. 4a). The GOME-2 and OMI  $VCD_{\text{tropo}}$   
330 show high values ( $>5 \times 10^{13} \text{ cm}^{-2}$ ) being maintained over the North Pole (Fig. 6). The MERRA-2 meteorological fields predict high surface winds in that area, which is covered by MYI (Fig. 7, bottom row). The  $VCD_{\text{tropo}}$  calculated by the FYI Snow simulation show little enhancement over the North Pole (Fig. 7), while the FYI+MYI Snow simulation better captures the magnitude and shape of the observed enhancement (Fig. 6), suggesting that blowing snow SSA emissions over MYI can be a significant source of BrO.

335 Overall, we find that the FYI+MYI Snow simulation captures reasonably well the spatial and temporal distribution of the observed  $VCD_{\text{tropo}}$ , as well as the spatial extent and frequency of BrO hotspots in March and April. In our simulation, blowing snow SSA emissions on MYI account for 20–30% of the  $VCD_{\text{tropo}}$  enhancement and could thus represent an important source of bromine activation. This is consistent with previous studies, which have suggested that, in addition to FYI, MYI could play a significant role in halogen activation. Gilman et al. (2010) used backtrajectories to calculate sea ice exposure of air masses  
340 observed during the International Chemistry Experiment in the Arctic Lower Troposphere (ICEALOT) ship-based study,



finding that exposure to both FYI and MYI was the best predictor of reduced O<sub>3</sub> levels. Choi et al. (2018) reported that the frequency of springtime BrO explosion events observed by OMI for 2005-2015 was strongly correlated with blowing snow SSA emissions over all sea ice, but they found little to no correlation when only FYI was considered as a source of blowing snow SSA. More generally, low O<sub>3</sub> and/or high BrO VCD<sub>tropo</sub> have been observed over and downwind of regions with a high  
345 fraction of MYI such as the Canadian Archipelago and the eastern Beaufort Sea (Bottenheim and Chan, 2006; Choi et al., 2012; Halfacre et al., 2014; Koo et al., 2012; Richter et al., 1998; Salawitch et al., 2010). In contrast to the generally good agreement in March and April, during May our simulation is systematically too low compared to GOME-2 VCD<sub>tropo</sub> (Figs. 1c and 4ab). In Section 5 we will discuss how the recycling of bromine deposition to snowpack in late spring could propagate blowing snow-induced halogen activation into May.

## 350 **4 Impact of blowing snow SSA on tropospheric O<sub>3</sub> over the Arctic**

### **4.1 Comparison to hourly surface O<sub>3</sub> observations**

We evaluate the ability of our simulations to capture ODEs via comparisons to hourly O<sub>3</sub> observations at several Arctic and sub-Arctic sites in 1 March - 31 May 2007 (Fig. 8). Observations at Utqiagvik and Alert show frequent occurrences of low O<sub>3</sub> mixing ratios (<10 ppbv), maintained for 1-7 days. These events are somewhat less frequent at Zeppelin. The FYI+MYI Snow  
355 simulation reproduces some of these events (such as the 8 – 18 April and 26 April – 10 May depletion events at Utqiagvik, the April events at Alert, many of the events at Zeppelin between late March to late April), but misses many others, in particular several March and late May events at Utqiagvik as well as the late April-May events at Alert.

Observed surface O<sub>3</sub> depletion events can be caused by advection of upwind O<sub>3</sub> poor air and/or local depletion (Bottenheim and Chan, 2006; Halfacre et al., 2014; Hopper et al., 1998; Jacobi et al., 2006; Simpson et al., 2007b, 2017). The rapid recovery  
360 of surface O<sub>3</sub> after a depletion event is often due to turbulent vertical mixing of O<sub>3</sub>-rich free tropospheric air down to the surface or to advection of O<sub>3</sub>-rich continental air (Bottenheim et al., 2009; Gong et al., 1997; Hopper et al., 1998; Jacobi et al., 2010; Moore et al., 2014; Morin et al., 2005). Furthermore, some ODEs can be highly localized. Using ice-tethered buoys over coastal regions and over the Arctic Ocean, Halfacre et al. (2014) found that large areas of the Arctic Ocean are partially depleted in O<sub>3</sub> during spring with local imbedded areas (~200-300 km) that are more depleted. Morin et al. (2005) report that  
365 dynamic conditions of the boundary layer near the shore can lead to very different surface O<sub>3</sub> behaviors observed at Alert compared to at a site over the sea ice 10 km away. Similar results were found by Jacobi et al. (2006), when comparing O<sub>3</sub> observations at Zeppelin Station and nearby (~100-200 km) ship-based measurements. The coarse resolution of our simulation (2°×2.5°) and the general difficulty of models in reproducing boundary layer depth and vertical mixing processes could thus be one explanation for our poor representation of ODEs at some of these surface sites. Another explanation is the lack of  
370 detailed chlorine chemistry in this version of GEOS-Chem. In particular, we do not consider acid displacement of Cl<sup>-</sup> in SSA which, together with bromine chemistry, could act to enhance ODEs. Finally, our simulation does not include local snowpack Br activation, which has been shown to lead to surface ODEs when the stable Arctic boundary layer is decoupled from

convective exchange with the free troposphere (Custard et al., 2017; Peterson et al., 2015, 2017; Pratt et al., 2013; Wang et al., 2019).

#### 375 4.2 Spatiotemporal distribution of O<sub>3</sub> depletion in GEOS-Chem

Fig. 9 shows the spatial distribution of monthly mean surface O<sub>3</sub> mixing ratios calculated in the FYI+MYI Snow simulation as well as the decrease in surface O<sub>3</sub> ( $\Delta O_3$ ) relative to the STD simulation for March and April 2007-2009. The lowest monthly mean O<sub>3</sub> mixing ratios are ~25 ppbv in March over the Canadian Archipelago and ~20-25 ppbv in April over the North Pole (Fig. 9a), with a spatial distribution corresponding to that of the simulated VCD<sub>tr<sub>o</sub>p<sub>o</sub></sub> (Fig. 2). Note that the low O<sub>3</sub> values simulated over Europe are due to NO<sub>x</sub> titration. The monthly mean values from the FYI+MYI Snow simulation are within better than 8 ppbv of surface observations at Utqiagvik (model/obs: March 28/22 ppbv, April 28/20 ppbv), Alert (model/obs: March 29/33 ppbv, April 27/27 ppbv), Zeppelin (model/obs: March 32/39 ppbv, April 27/31 ppbv), and Elk Island (model/obs: March 28/36 ppbv, April 35/38 ppbv). The mean values of surface  $\Delta O_3$  increase from 4-8 ppbv (15-30% depletion relative to the STD simulation) in March, to 8-14 ppbv (30-40%) in April, when sufficient sunlight is available to drive photochemistry at high latitudes (Fig. 9b). Poleward of 60° N, the FYI+MYI Snow simulation displays a pan-Arctic O<sub>3</sub> decrease of 3.7 ppbv (11%) in March and 8.3 ppbv (23%) relative to the STD simulation.

We define the frequency of ODEs as the percent of time when more than 20 ppbv O<sub>3</sub> is lost due to blowing snow ( $\Delta O_3 > 20$  ppbv, Fig. 9c). Applying this definition to the FYI+MYI Snow simulation, we find that ODEs occur up to 1-5% of the time in March, increasing to up to 15-25% of the time in April as sunlight extends to higher latitudes (Fig. 9c). Using an ODE definition including more moderate events ( $\Delta O_3 > 10$  ppbv), we find that poleward of 70° N ODEs occur 20-60% of the time in April (Fig. 9d).

The decrease in surface O<sub>3</sub> due to blowing snow SSA is 60% larger in the FYI+MYI Snow simulation compared to the FYI Snow simulation (Fig. 10a). The timeseries of pan-Arctic daily mean  $\Delta O_3$  for the FYI+MYI Snow simulation at different altitudes shows that  $\Delta O_3$  is uniform in the bottom 500 m altitude and extends to 1000 m altitude, where the model-calculated  $\Delta O_3$  is about half of the surface  $\Delta O_3$ . At 2000 m,  $\Delta O_3$  is generally below 4 ppbv (Fig. 10b). This is consistent with ozonesonde profiles, which indicate that depletion events are confined to the lowest 1000 m with an average height of the top of the layer at 500 m (Hopper et al., 1998; Oltmans et al., 2012; Tarasick and Bottenheim, 2002). We find that the Arctic tropospheric O<sub>3</sub> burden (>60° N) decreases by 3% in March and 6% in April in the FYI+MYI simulation relative to the STD simulation.

In our simulation, the timing of the maximum  $\Delta O_3$  takes place 4-5 days after the maximum in the blowing snow SSA burden (Figs. 4c and 10b), reflecting increasing O<sub>3</sub> loss as BrO concentrations increase. This is illustrated during the March 25-30 2007 event: the pan-Arctic blowing snow SSA burden and VCD<sub>tr<sub>o</sub>p<sub>o</sub></sub> reach their maximum value on 27 March, while the lowest values for surface O<sub>3</sub> occur 4 days later on 31 March (Fig. 5, bottom row). By 29-30 March the blowing snow SSA burden is back to low values over the Arctic, but the VCD<sub>tr<sub>o</sub>p<sub>o</sub></sub> are still elevated and low surface O<sub>3</sub> mixing ratios are predicted throughout



the Arctic. This can be explained by the different lifetimes of these species over the springtime Arctic: ~1 day for SSA, 4-7  
405 days for the Br<sub>y</sub> family (consisting of all the gas-phase inorganic bromine species), and 30-40 days for O<sub>3</sub>.

#### 4.3 Transport of O<sub>3</sub>-depleted Arctic air to lower latitudes

In some instances, large-scale blowing snow induced bromine activation can influence surface O<sub>3</sub> in sub-Arctic regions. The  
large BrO explosion and associated O<sub>3</sub> depletion from late March to early April 2007 (Figs. 4ab) illustrates this. The green  
shading in Fig. 8 corresponds to the times when this BrO explosion is transported to the three Arctic sites and several sub-  
410 Arctic sites. The GEOS-Chem FYI+MYI Snow simulation shows transport of O<sub>3</sub>-depleted air to Utqiagvik on 25-27 March  
2007, with observations showing much stronger depletion than the model (Fig. 8a). This air mass is observed at Alert on 31  
March-4 April and the eastern edge of that O<sub>3</sub>-depleted air reaches Zeppelin around the same time (Figs. 8bc). The O<sub>3</sub>-depleted  
air is then transported southward over Hudson Bay and towards the northeastern United States during 4-9 April (Figs. 5 and  
11), mostly below 1,000 m altitude. At the synoptic level, a surface low was sweeping across the United States from the central  
415 Rockies to the Great Lakes in early April, bringing sub-freezing Arctic air behind its cold front with record-breaking cold  
temperatures being measured over the Central Plains and much of the Southeast (NOAA/USDA, 2008).

Between 2 April and 7 April, observed O<sub>3</sub> mixing ratios at 5 sub-Arctic sites from Elk Island (53.7° N) to Woodstock (43.9°  
N), decreased from background levels of 40 ppbv down to 10-20 ppbv (Fig. 8). This decrease is reproduced by both the FYI  
and FYI+MYI Snow simulations, which show that enhanced VCD<sub>tropo</sub> and blowing snow SSA were transported over Hudson  
420 Bay towards the Great Lakes region (Fig. 11a-c). By the time the polar air mass reached the Northeast United States, the BrO  
VCD<sub>tropo</sub> were back to normal levels, but because of its longer lifetime, O<sub>3</sub> was 5-10 ppbv lower than background mixing ratios  
as shown by the FYI+MYI Snow simulation (Fig. 11d). This decrease in surface O<sub>3</sub> was observed at NAPS and CASTNET  
surface sites throughout Southeast Canada and the Northeast United States (Fig. 11d).

Ridley et al. (2007) discuss a similar transport event of O<sub>3</sub>-depleted air from the Canadian Archipelago and nearby Arctic  
425 Ocean to Hudson Bay observed in April 2000 during the Tropospheric Ozone Production about the Spring Equinox (TOPSE)  
aircraft campaign. During that event, O<sub>3</sub> levels were reduced from 30-40 ppbv down to values as low as 0.5 ppbv in a large  
area below 500 m altitude. Over the 3-year period that we examined, our simulations suggest that transport of O<sub>3</sub>-depleted air  
( $\Delta O_3 > 20$  ppbv) to Hudson Bay occurs 1-10% of the time (0.3-3 days/month) in March and in April (Fig. 9c), with transport of  
more moderate O<sub>3</sub> depletion ( $\Delta O_3 > 10$  ppbv) taking place 10-20% of the time (3-6 days/month). In March-April 2008,  
430 measurements in the ice-free North Atlantic showed transport of O<sub>3</sub>-poor air from the Arctic basin down to 52° N during the  
ICEALOT cruise (Gilman et al., 2010): a 13 ppbv decrease in O<sub>3</sub> was accompanied by a simultaneous decrease in the acetylene  
to benzene ratio indicating exposure to halogen oxidation. In our simulation, we find that transport of air mass with  $\Delta O_3 > 10$   
ppbv air to sub-Arctic latitudes occurs with a frequency of 5-10% (1.5-3 days/month) down to 50° N and 1-5% (< 1.5  
days/month) down to 40° N (Fig. 9d) and that this transport appears to be favored over Hudson Bay extending to the Northeast



435 United States and the Western Atlantic off Nova Scotia (Fig. 9d). At mid-latitudes (30°-60° N), the FYI+MYI Snow simulation results in a mean surface O<sub>3</sub> decrease of 1.2 ppbv (3.3%) in March-April, thus a small but non-negligible contribution.

### 5 Atmospheric deposition on snow as a source of salinity and bromide

We use the results of the FYI Snow simulation to examine the potential role of atmospheric deposition of SSA and gas-phase Br<sub>y</sub> as a source of salinity and bromide to snow on sea ice. Snow composition profiles indicate that atmospheric deposition can sometimes be a more important source of salinity and Br<sup>-</sup> than upward brine migration, especially over MYI and in deep snowpack (>10-17 cm) over FYI (Domine et al., 2004; Krnavek et al., 2012; Peterson et al., 2019). Fig. 12a shows the evolution of deposition on sea ice (poleward of 60° N) for Na<sup>+</sup> and for total bromine (sum of particulate Br<sup>-</sup> and gas-phase Br<sub>y</sub>) in the GEOS-Chem FYI Snow simulation. Between September and May, deposition of Na<sup>+</sup> on snow-covered sea ice has an average daily value of 2.2×10<sup>6</sup> kg/day. For total bromine (SSA Br<sup>-</sup> and gas-phase Br<sub>y</sub>), the average daily value is 6.7×10<sup>4</sup> kg/day. Open ocean SSA and blowing snow SSA account nearly equally to these deposition fluxes.

From October to March the deposition of total bromine tracks that of Na<sup>+</sup> as both are derived from SSA (Fig. 12a). Starting in late March, deposition of Na<sup>+</sup> decreases rapidly following the decrease in blowing snow SSA emissions (Fig. 4c) due to slower wind speeds and warmer temperatures (Huang and Jaeglé, 2017). Deposition of bromine remains high for another month, however, as once bromine activation starts in early spring a large fraction of Br deposition is in the form of Br<sub>y</sub>, in particular HBr which is the end product of reactive bromine chemistry. In April-May, deposition of Br<sub>y</sub> accounts for more than half of total bromine deposition (Fig. 12a). As Br<sub>y</sub> has a longer lifetime against deposition than particulate Br<sup>-</sup> and Na<sup>+</sup>, Br<sub>y</sub> shows a more gradual decrease in deposition in April and May. This behavior is consistent with observations of freshly fallen snow at Alert during the Network on Climate and Aerosols Research (NETCARE) campaign, showing a broad spring peak in Br<sup>-</sup> deposition between late March and late May (Macdonald et al., 2017). A similar increase of Br<sup>-</sup> in snowfall samples after polar sunrise was reported at Alert by Toom-Sauntry & Barrie (2002). Furthermore, Spolaor et al. (2013, 2014) found a strong seasonal variability of Br<sup>-</sup> in polar firn at both Arctic and Antarctic sites, with greater Br<sup>-</sup> values in spring/summer compared to winter.

We derive the Br<sup>-</sup>/Na<sup>+</sup> enrichment factor (EF) of deposition relative to seawater composition (which has a Br<sup>-</sup>/Na<sup>+</sup> mass ratio of 0.00625 g/g):

$$EF = \left( \frac{[Br^- + Br_y]}{[Na^+]} \right)_{deposition} / \left( \frac{[Br^-]}{[Na^+]} \right)_{seawater}$$

EF is calculated for each day and each model grid box and then surface-area weighted over sea-ice to obtain the daily timeseries of pan-Arctic EF shown in Fig. 12b. The model-calculated EF is fairly constant between September and February, with values of ~3-10. Starting in late March, the divergence between bromine and Na<sup>+</sup> deposition leads to an increase in EF to a value of



14 by the end of April, followed by a more rapid increase in May up to a value of  $\sim 60$ . In May, the mean model-calculated EF  
465 of deposition is 30. These values are remarkably consistent with the EF measured during the NETCARE campaign at Alert  
(Macdonald et al., 2017), as shown in Fig. 12b, with mean observed values of  $EF \sim 7$  between September and February, 15 in  
April, and 25 in May. Similarly, Toom-Sauntry & Barrie (2002) observed an increase in EF from low enrichment in the dark  
winter months (median 1.5-5) to a large enrichment (median 20-72) after polar sunrise. In our simulation of the blowing snow  
SSA, we assumed a constant value of  $EF=5$ , which appears to underestimate the EF of snow in May by a factor of 2-3. This  
470 springtime increase in the enrichment of  $Br^-$  in snow over sea ice could thus act to propagate bromine explosions and ODEs  
into May, and could explain our systematic underestimate in observed  $VCD_{tropo}$  in May. Using a one-dimensional model, Piot  
& von Glasow (2008) compared modeled and observed deposition on snow at Utqiagvik, demonstrating that 75% of deposited  
bromine may be re-emitted into the gas phase as  $Br_2$  or  $BrCl$ . They proposed that cycles of deposition and re-emissions of  
bromine from snow could thus result in a “leap-frogging” process and explain observations of progressive  $Br^-$  enrichment of  
475 snow in coastal Arctic regions with distance inland (Simpson et al., 2005). Our simulations suggest that a similar mechanism  
could occur in snow over sea ice via repeated cycles of blowing snow SSA sublimation and deposition of reactive bromine.  
These cycles could thus be important in propagating bromine explosions in space (onto coastal snow-covered land) as well as  
in time on sea ice (into late spring).

We now use the model-simulated  $Na^+$  deposition (FYI Snow simulations) on the sea ice covered Arctic ocean to estimate the  
480 salinity of surface snow due to SSA deposition. In our simulation, wet and dry deposition contribute nearly equally to SSA  
deposition. Over the 3-month period between February 1 and April 31, the cumulative SSA deposition on Arctic sea ice is  
 $10.3 \times 10^8$  kg SSA in the FYI snow simulation. Snow depth on sea ice is not calculated in MERRA-2 and reanalyses tend to  
have highly uncertain solid precipitation in polar regions (Lindsay et al., 2014; Uotila et al., 2019), so we rely on the limited  
observations available. The Warren et al. (1999) climatology, which is based on 1954-1991 Soviet drifting stations  
485 measurements on Arctic MYI, shows that snow depth increases from 29.7 cm to 34.4 cm during these three months, resulting  
in 4.7 cm snow accumulation. This is likely an overestimate of the pan-Arctic snow depth because MYI tends to have thicker  
snow and these measurements represent conditions from past decades (Kwok and Cunningham, 2008). Indeed, the 2003-2008  
AMSR-E snow depth retrievals over FYI compiled by Zygmontowska et al. (2014) show much lower snow depths, increasing  
from 18 cm in February to 19 cm in in May, thus a 1 cm snow accumulation. These two estimates in snow accumulation result  
490 in pan-Arctic snow accumulations of  $10^{14}$  kg to  $2.2 \cdot 10^{13}$  kg (sea ice extent is 11 million  $km^2$  and fresh snow has a density of  
 $200 \text{ kg/m}^3$ ), which combined with our calculated SSA deposition results in a mean salinity of 0.01 psu - 0.04 psu of surface  
snow due to deposition. Fig. 12c shows the spatial distribution of snow salinity due to SSA deposition assuming a 1 cm snow  
accumulation. High salinities ( $>0.05$  psu) are calculated in sea-ice regions near the ice-free open ocean in the North Atlantic  
and Bering Sea, with lower values on FYI (0.02-0.05 psu) and MYI (0.01-0.03 psu). Deposition could thus represent a  
495 significant source of salinity of surface snow on MYI by the time sunrise reaches high latitudes, and it could also be important  
for FYI as the snow gets deeper in late spring (Petty et al., 2018; Warren et al., 1999).



## 6 Conclusions

We used the GEOS-Chem model to examine the impact of blowing snow SSA on bromine chemistry and O<sub>3</sub> during Arctic spring. We conduct two blowing snow simulations assuming a 0.1 psu surface snow salinity on FYI and different salinities on MYI (0.01 psu for the FYI Snow simulation and 0.05 psu for the FYI+MYI Snow simulation). We evaluated these simulations against satellite observations of VCD<sub>tropo</sub> from GOME-2 and OMI for 2007-2009. We found that our simulations reproduce the spatio-temporal distribution of the observed VCD<sub>tropo</sub> to within -12% to +17% in March and April and capture the spatial extent as well as the frequency of most large BrO hotspots in March and April. The FYI+MYI Snow simulation captured observed BrO hotspots over the North Pole better than the FYI Snow simulation. In our FYI+MYI simulation, blowing snow SSA emissions on MYI account for 20-30% of the VCD<sub>tropo</sub> enhancement and thus represent a significant source of bromine activation. In the past, the search for halogen-containing substrates in polar regions has often focused on the role of frost flowers, newly formed sea ice, and FYI snowpack on polar halogen activation, because of the higher salinity of these surfaces and of their snowpack relative to MYI (e.g., Abbatt et al., 2012; Jacobi et al., 2006; Kaleschke et al., 2004; Simpson et al., 2007a; Yang et al., 2010). Our simulations suggest that the role of MYI in halogen activation is likely also important, consistent with results of several studies (Choi et al., 2018; Gilman et al., 2010; Halfacre et al., 2014; Peterson et al., 2019).

Inclusion of blowing snow SSA in our FYI+MYI Snow simulation results in a pan-Arctic decrease in surface O<sub>3</sub> of 3.7 ppbv (11%) in March and 8.3 ppbv (23%) in April. As most of this decrease is confined to altitudes below 1000 m, the Arctic tropospheric O<sub>3</sub> burden decreases by only 3-6% in March-April. Compared to surface O<sub>3</sub> observations at coastal Arctic sites, we find that our simulation often underestimates the magnitude of observed ODEs and sometimes misses observed ODEs. The simulation does reproduce an event of low O<sub>3</sub> Arctic air transport down to 40° N, which was observed at multiple sub-Arctic surface sites. We estimate that these transport events occur in March and April with a frequency of 1.5-3 days/month down to 50° N and < 1.5 days/month down to 40° N.

The mixed success of our simulation at capturing the rapid O<sub>3</sub> variations observed at coastal Arctic sites could be related to the coarse resolution of the model, the difficulty in simulating Arctic boundary layer exchange processes, the lack of detailed chlorine chemistry, or the fact that we did not include direct halogen activation by snowpack chemistry. The influence of snowpack bromine chemistry was examined in two previous modeling studies (Falk and Sinnhuber, 2018; Toyota et al., 2011). These studies displayed remarkable agreement with surface O<sub>3</sub> observations, in particular at Alert and Utqiagvik, and reproduced the synoptic variability in GOME VCD. However, both studies systematically underestimated the magnitude of observed BrO VCDs by at least a factor of 2. These results together with our GEOS-Chem blowing snow simulations suggest that most of the VCD<sub>tropo</sub> enhancements observed by satellites can be explained by blowing snow SSA leading to BrO enhancements over larger vertical (0-2 km) and horizontal scales (pan-Arctic) than snowpack chemistry, but that in the shallow boundary layer (~50-250 m) over the springtime Arctic direct snowpack halogen activation dominates Br release and is responsible for the most severe ODEs. Indeed, some of the strongest ODEs observed at the surface seem to occur when the



530 stable Arctic boundary layer is decoupled from convective exchange with the free troposphere (Moore et al., 2014; Seabrook et al., 2011; Wang et al., 2019). Furthermore as boundary layer O<sub>3</sub> reaches very low levels (<4 ppbv), BrO production via R3 is suppressed and atomic Br becomes much more abundant than BrO (Neuman et al., 2010; Wang et al., 2019), such that satellites would not necessarily observe co-located O<sub>3</sub> depletion and enhanced BrO VCD<sub>tropo</sub>.

535 Based on our analysis of the seasonal variation in modeled deposition of Na<sup>+</sup> and bromine on snow, we propose that the progressive enrichment of bromine in deposition onto sea ice could help propagate blowing snow SSA bromine activation into May, even as the magnitude of blowing snow SSA emissions starts to decrease. The increase of our calculated springtime enrichment of Arctic snow Br<sup>-</sup> is similar to observations showing a peak after polar sunrise (Macdonald et al., 2017; Spolaor et al., 2013; Toom-Saunty and Barrie, 2002). Recycling of deposited bromine was previously proposed in the context of frost flower/snowpack activation followed by inland transport over coastal regions (Domine et al., 2004; Piot and Glasow, 2008; Simpson et al., 2005, 2007b). We propose that a similar mechanism takes place over sea ice and snow-covered coastal regions with blowing snow SSA, enabling BrO explosions to propagate spatially and to last longer into late spring. We also show that SSA deposition to surface snow in winter-spring, when snow accumulation on sea ice is at its minimum, could account for 0.01-0.03 psu on MYI, 0.02-0.05 psu on FYI, and >0.1 psu on sea ice areas close to the open ocean. While upward migration of brine from sea ice is the main source of salinity in surface snow over FYI with shallow snowpack, atmospheric deposition could thus be the dominant source in surface snow over the less saline MYI sea ice and could play an important role in late spring as the snowpack deepens over FYI (Domine et al., 2004; Krnavek et al., 2012; Peterson et al., 2019).

Our simulations did not include 2-way coupling between snowpack composition and atmospheric deposition, did not consider blowing snow on coastal snow enriched in Br<sup>-</sup>, and did not incorporate direct snowpack halogen activation. Incorporating these coupled processes together with blowing snow SSA promises to yield further insights into the mechanisms leading to polar bromine activation and ODEs.

550 **Data availability.** The GOME-2 VCD<sub>tropo</sub> data is available upon request from Nicolas Theys (Nicolas.Theys@aeronomie.be). The OMI VCD<sub>tropo</sub> data is available upon request from Sungyeon Choi (sungyeon.choi@nasa.gov). The surface O<sub>3</sub> observations are available from the Observations from these surface monitoring sites are obtained through the World Data Centre for Reactive Gases (WDCRG, <http://ebas.nilu.no>), CAPMoN (<http://donnees.ec.gc.ca/data/air/monitor/monitoring-of-atmospheric-gases/ground-level-ozone/>), NAPS (<https://www.canada.ca/en/environment-climate-change/services/air-pollution.html>), and CASTNET (<https://www.epa.gov/castnet/castnet-ozone-monitoring>) data repositories. The GEOS-Chem simulations are available upon request from the corresponding author (jaegle@uw.edu).



**Author contributions.** JH and LJ designed the study, conducted the GEOS-Chem simulations, and analysed the results. QC, BA, TS, ME contributed to model development. NT and SC provided the tropospheric BrO columns for GOME-2 and OMI. JH wrote the paper with significant contributions from LJ. All the authors contributed to editing the paper.

560 **Competing interests.** The authors declare that they have no conflict of interest.

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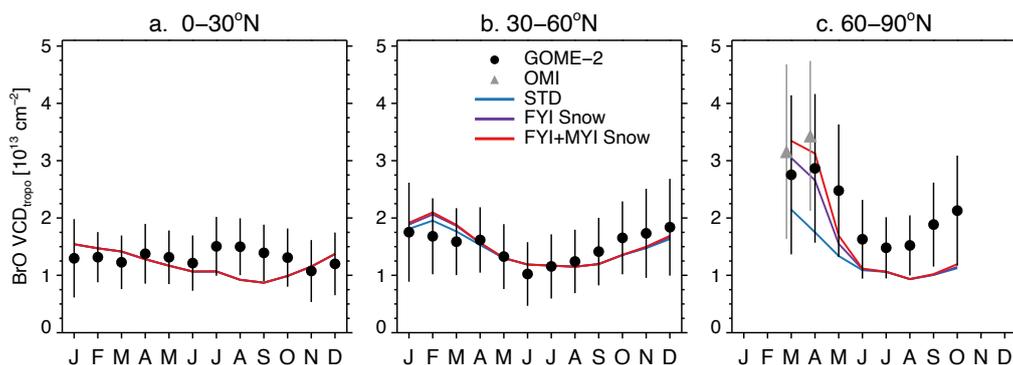
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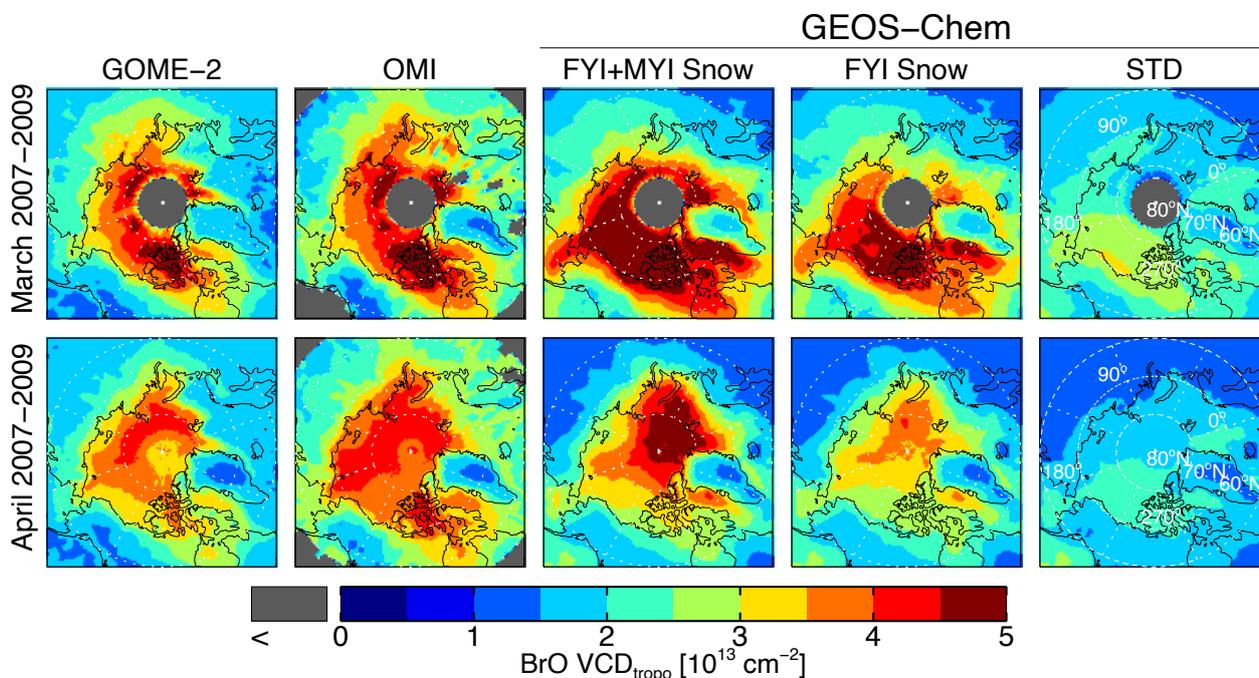


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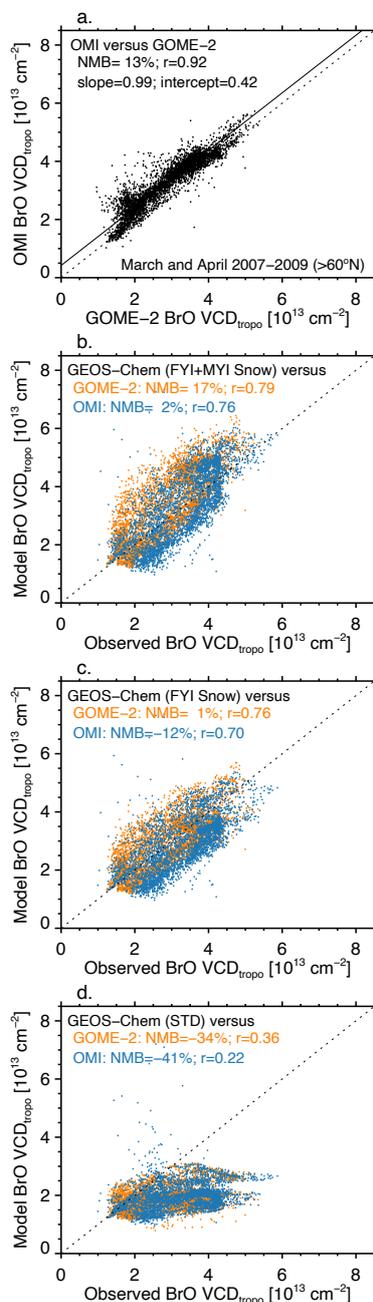


895 **Figure 1:** Seasonal variation in 2007-2009 monthly mean tropospheric BrO vertical column densities ( $VCD_{\text{tropo}}$  in units of  $10^{13}$  molecules  $\text{cm}^{-2}$ ) observed by GOME-2 (black circles) and OMI (gray triangles in panel c for April and May), and simulated with GEOS-Chem over (a) 0–30° N, (b) 30–60° N and (c) 60–90° N. The three GEOS-Chem simulations shown are the standard simulation (STD, blue line) and two blowing snow simulations (FYI Snow, purple line and FYI+MYI Snow, red line). The black and gray error bars represent 1 standard deviation about the monthly mean GOME-2 and OMI  $VCD_{\text{tropo}}$ . For each latitude bin, we only show means for months where at least 70% of the surface area has valid satellite observations.

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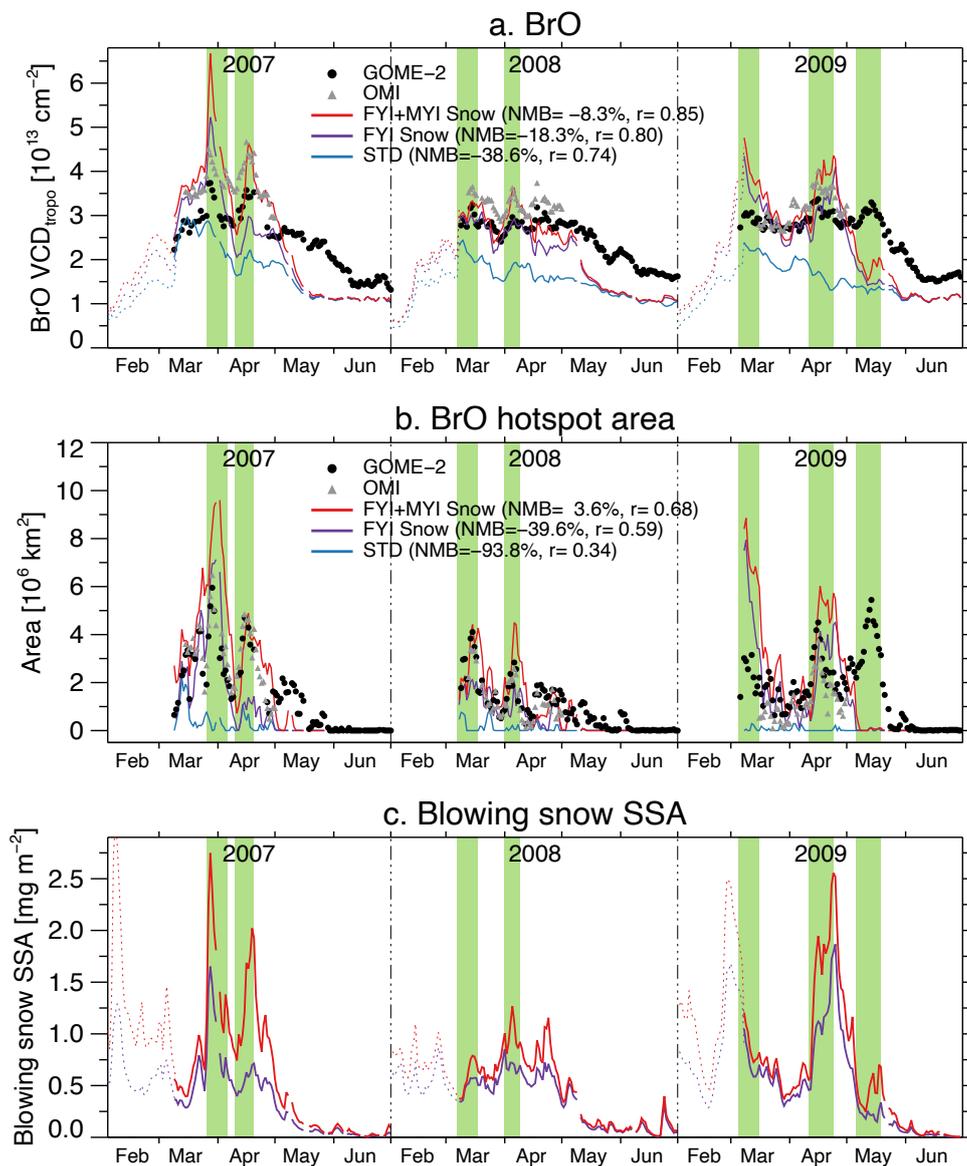
**Figure 2:** Spatial distribution of  $VCD_{\text{tropo}}$  in March (top row) and April (bottom row) 2007–2009. Satellite retrievals from GOME-2 and OMI are compared to the GEOS-Chem FYI+MYI Snow, FYI Snow, and STD simulations.



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Figure 3: (a) Scatterplot of GOME-2 and OMI monthly mean VCD<sub>тропо</sub> for March and April 2007-2009 poleward of 60° N. (b) Scatterplot of VCD<sub>тропо</sub> calculated with the FYI+MYI Snow simulation and retrieved by GOME-2 (orange circles) and OMI (blue circles). (c) Same as (b) but for the FYI Snow simulation. (d) Same as (b) but for the STD simulation. The Normalized Mean Bias (NMB, see definition in section 3.1), Pearson correlation coefficient, and slope are shown in each panel. For panel (b), the slope and intercept of the reduced-major-axis regression line (solid line) are also indicated. The dashed line corresponds to the 1:1 line.

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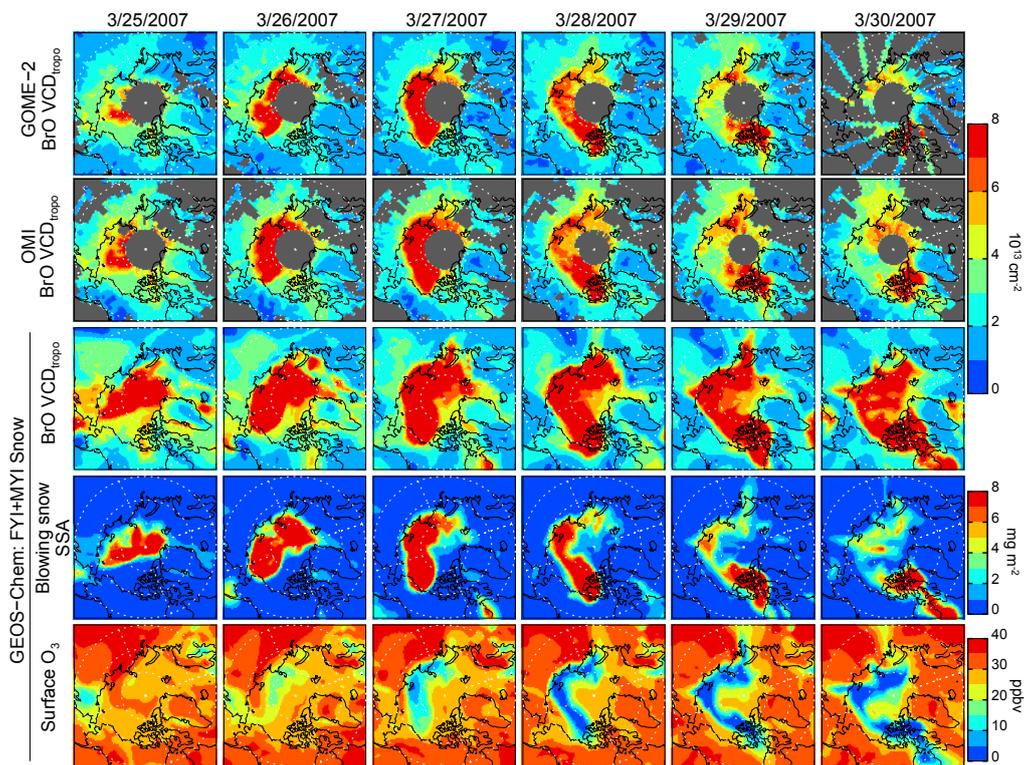


915 **Figure 4:** Timeseries of daily mean  $\text{BrO VCD}_{\text{tropo}}$ , tropospheric BrO hotspots area, and blowing snow SSA burden over high latitudes ( $>60^\circ \text{ N}$ ) between February and June for 2007–2009. (a) Timeseries of daily  $\text{VCD}_{\text{tropo}}$  (in  $10^{13} \text{ cm}^{-2}$ ) averaged poleward of  $60^\circ \text{ N}$  for GOME-2 (black circles), OMI (gray triangles) and simulated with GEOS-Chem (FYI+MYI Snow: red, FYI Snow: purple, STD: blue). (b) The daily area extent of BrO hotspots (in units of  $10^6 \text{ km}^2$ ) for GOME-2, OMI, and GEOS-Chem. See section 3.2 for the definition of BrO hotspot areas. Note that the total area poleward of  $60^\circ \text{ N}$  is  $34.2 \cdot 10^6 \text{ km}^2$ . (c) Timeseries of daily mean blowing snow SSA burden ( $\text{mg m}^{-2}$ ) for the FYI+MYI Snow and FYI Snow simulations. The blowing snow SSA burden is obtained as the difference between the blowing snow and STD simulations. The normalized mean bias (NMB) and Pearson correlation coefficients displayed in panels (a) and (b) are relative to the combined GOME-2 and OMI timeseries. The events highlighted in light green are defined as periods when GOME-2 or OMI BrO hotspots cover more than  $2 \cdot 10^6 \text{ km}^2$  for longer than 5 days. We only show days where valid satellite observations are available over at least 70% of the surface area poleward of  $60^\circ \text{ N}$ . The dashed lines in (a) and (c) correspond to the GEOS-Chem daily means poleward of  $60^\circ \text{ N}$  for days with less than 70% of valid observations.

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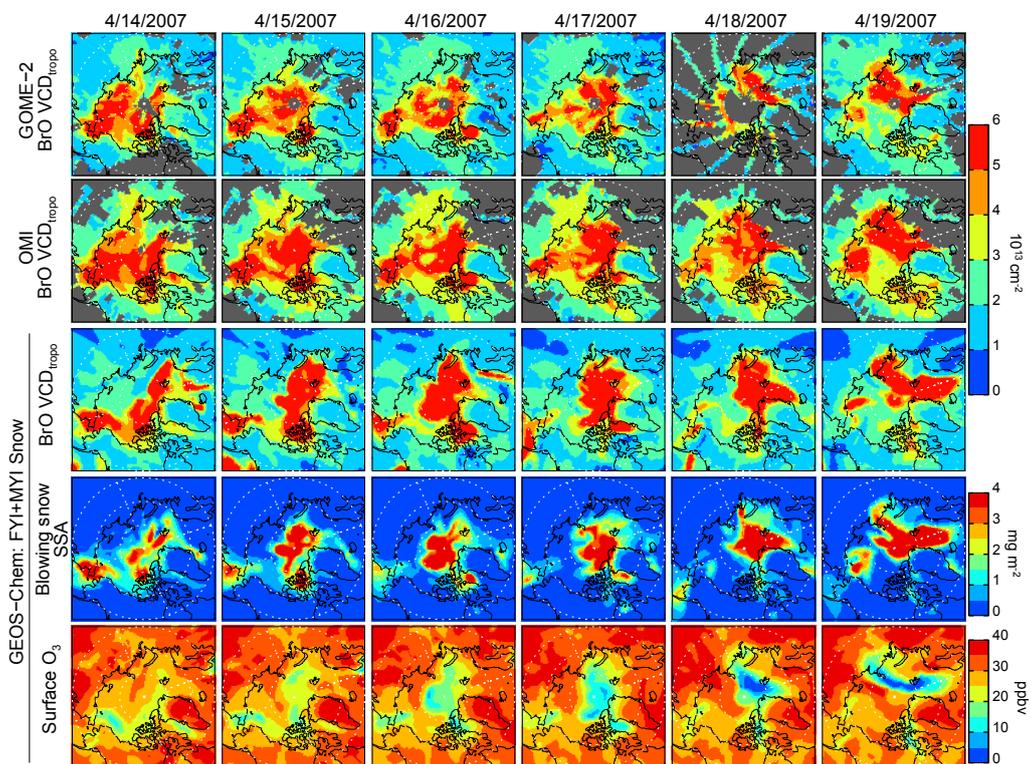


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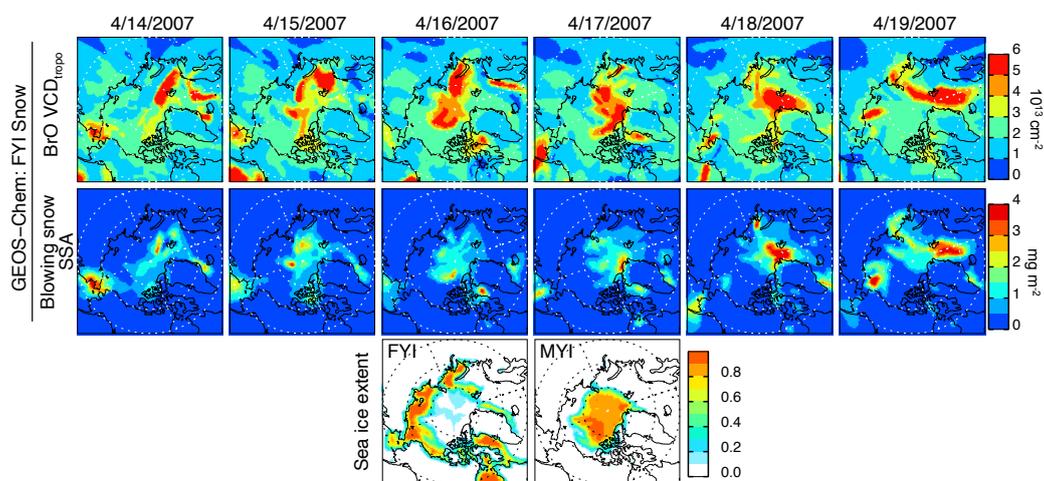


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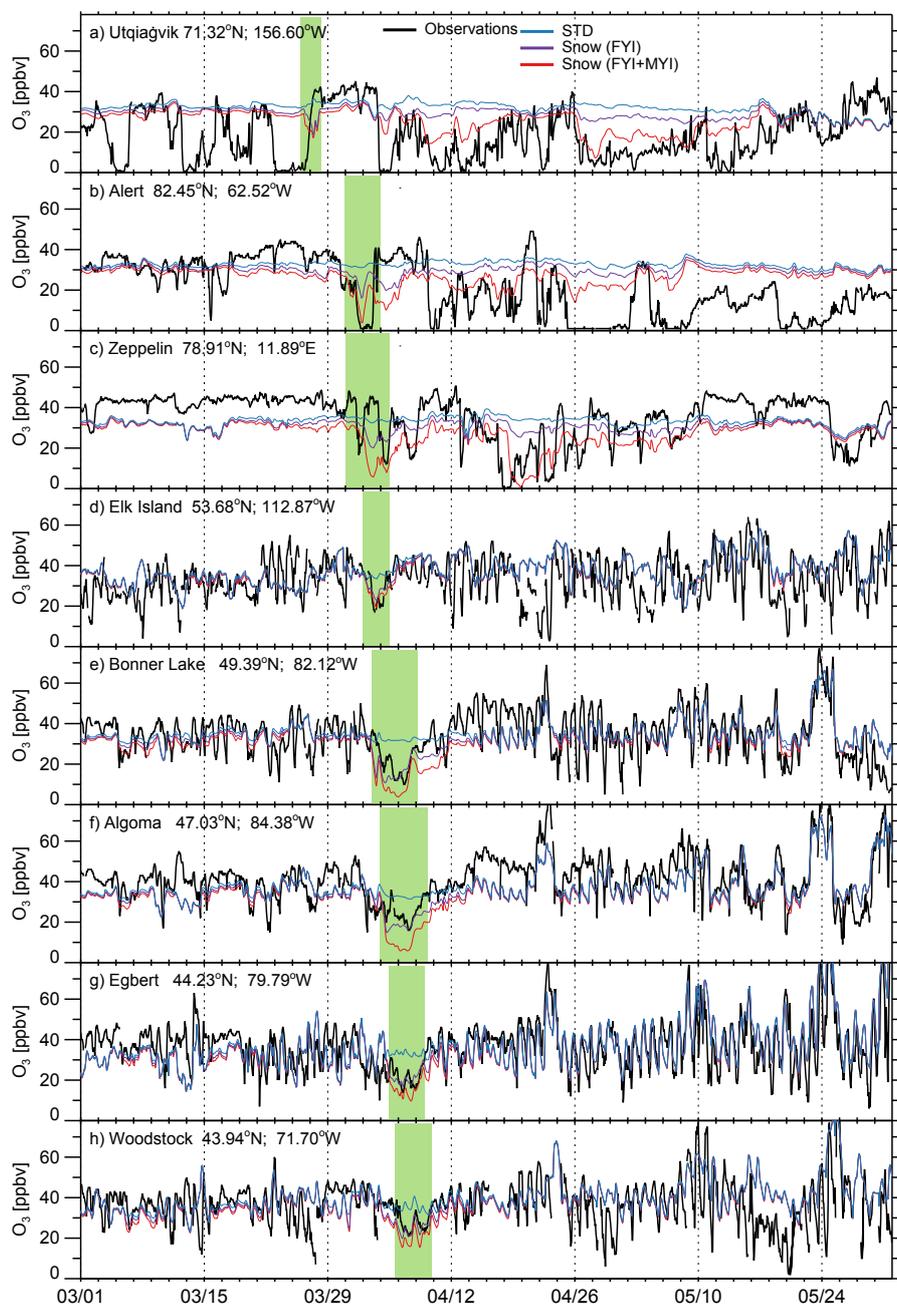
Figure 5: Daily 25-30 March 2007 distribution of BrO VCD<sub>tropo</sub> for GOME-2, OMI and GEOS-Chem (FYI+MYI Snow simulation). Also shown are the blowing snow SSA mass burdens ( $\text{mg m}^{-2}$ ) and surface  $\text{O}_3$  mixing ratios (ppbv) for the FYI+MYI Snow simulation. The gray areas in the first two rows correspond to regions with no BrO retrievals. Note that polar sunrise occurred equatorward of  $80^\circ \text{ N}$ .



935 **Figure 6: Daily 14-19 April 2007 distribution of BrO VCD<sub>tropo</sub> from GOME-2, OMI and GEOS-Chem (FYI+MYI Snow simulation). Also shown are the blowing snow SSA burdens ( $\text{mg m}^{-2}$ ) and surface  $\text{O}_3$  mixing ratios (ppbv) in the FYI+MYI Snow simulation.**



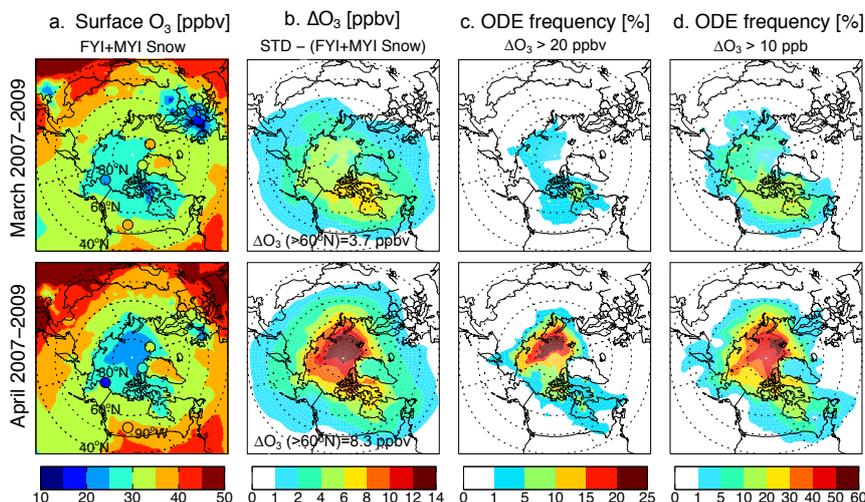
940 **Figure 7: Daily distribution of BrO VCD<sub>tropo</sub> and blowing snow SSA burden from the GEOS-Chem FYI Snow simulation for 14-19 April 2007. The bottom row shows the distribution of sea ice extent for FYI and MYI.**



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**Figure 8: Timeseries of hourly surface O<sub>3</sub> (ppbv) at three Arctic sites (a,b,c) and 5 sub-arctic sites (d-h) for March 1-May 31 2007: a) Utqiagvik (Barrow), Alaska, USA, b) Alert, Nunavut, Canada, c) Zeppelin, Spitsbergen, Norway, d) Elk Island, Alberta, Canada, e) Bonner Lake, Ontario, Canada, f) Algoma, Ontario, Canada, g) Egbert, Ontario, Canada h) Woodstock, New Hampshire, USA. In situ observations are shown with the black lines, while the GEOS-Chem STD, FYI Snow, and FYI+MYI Snow simulations are shown in blue, purple, and red, respectively. The shaded green areas corresponding to the times when the large BrO hotspot from 26 March - 4 April 2007 is predicted to be observed at the 3 Arctic sites and then transported to the sub-arctic sites.**

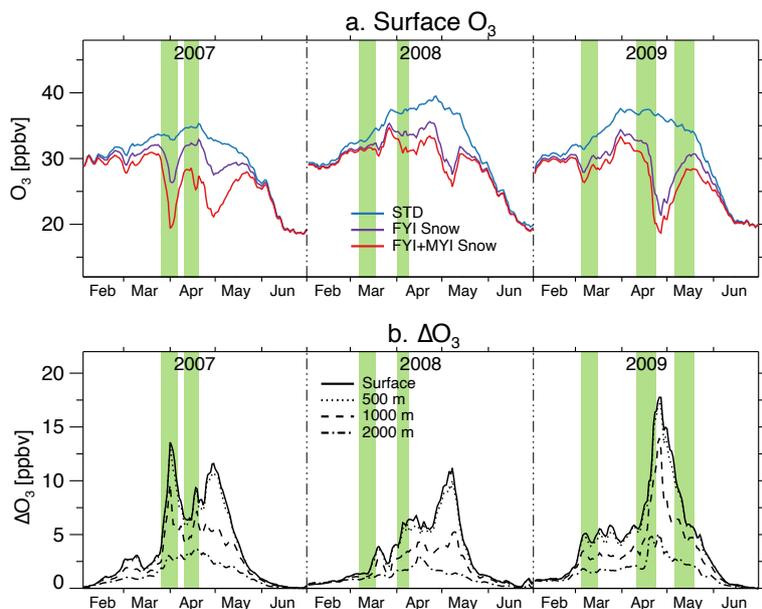
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**Figure 9:** Monthly mean distribution of surface  $O_3$  simulated with GEOS-Chem for March (top row) and April (bottom row) 2007-2009. Column a) Surface  $O_3$  mixing ratios in the GEOS-Chem FYI+MYI Snow simulation. The circles are color-coded by monthly mean surface  $O_3$  observed at 3 Arctic (Utqiagvik, Alert, Zeppelin) and 1 sub-Arctic site (Elk Island). Column b) Decrease in surface  $O_3$  ( $\Delta O_3$ , ppbv) due to blowing snow obtained as the difference between the STD and FYI+MYI Snow simulations. Column c): Occurrence frequency of ODEs (in %), defined as the percent of time that more than 20 ppbv of  $O_3$  is lost due to blowing snow ( $\Delta O_3 > 20$  ppbv). Column d): Occurrence frequency of ODEs, defined as  $\Delta O_3 > 10$  ppbv.

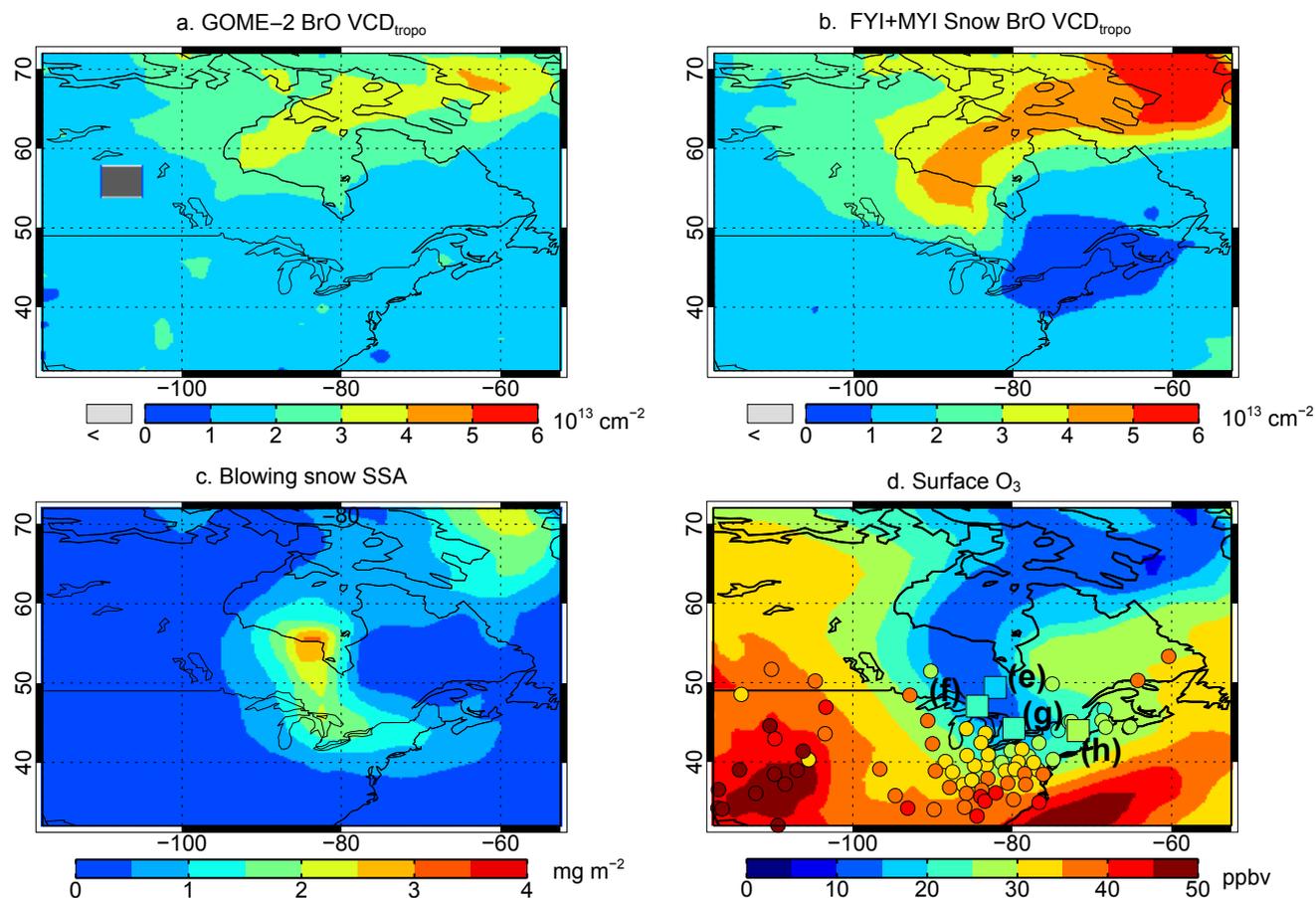
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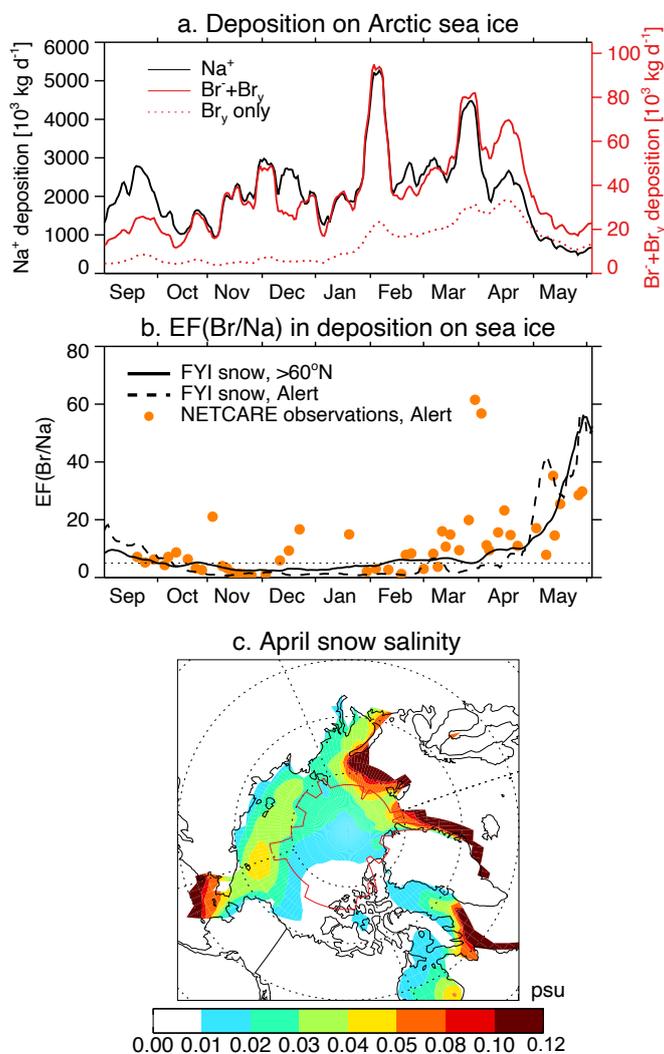
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**Figure 10:** Timeseries of daily mean surface  $O_3$  and decrease in surface  $O_3$  ( $\Delta O_3$ ) due to blowing snow at high latitudes ( $>60^\circ N$ ) between February and June 2007-2009. (a) Daily mean surface  $O_3$  mixing ratios averaged poleward of  $60^\circ N$  for the STD (blue line), FYI Snow (purple) and FYI+MYI Snow (red line) GEOS-Chem simulations. (b) Daily mean  $\Delta O_3$ , obtained as the difference between the FYI+MYI Snow and STD simulations at the surface (solid line), 500 m altitude (dotted line), 1000 m altitude (dashed line), and 2000 m altitude (dashed-dotted line). The events highlighted in light green are defined as periods when GOME-2 or OMI BrO hotspots cover more than  $2 \cdot 10^6$  km<sup>2</sup> for longer than 5 days.

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975 **Figure 11: Mean 5-9 April 2007 VCD<sub>tropo</sub> for (a) GOME-2 and (b) the GEOS-Chem FYI+MYI Snow simulation. Mean 5-9 April 2007 (c) blowing snow SSA burden and (d) surface O<sub>3</sub> mixing ratios for April 5-9 2007 from the GEOS-Chem FYI+MYI Snow simulation. Surface O<sub>3</sub> observations are shown as color-coded symbols in panel (d), with the squares correspond to the sub-Arctic sites from Fig. 8. The letter numbering (e-h) corresponding to the panel numbers in Fig. 8.**



980 **Figure 12: Modeled evolution of deposition and Br/Na enrichment factor on sea ice (>60° N) for 2007 calculated with the FYI Snow simulation. (a) Total deposition on sea ice for Na<sup>+</sup> (black line, left axis), Br<sup>-</sup>+Br<sub>y</sub> (red line, right axis), and Br<sub>y</sub> only (dashed red line, right axis) in units of 10<sup>3</sup> kg/day. (b) Modeled Br/Na enrichment factor EF (black line) relative to seawater in deposition over sea ice, see section 5 for the definition of EF. The orange circles show the enrichment factors observed during the NETCARE campaign at Alert, Nunavut (Macdonald et al., 2017). The dashed black line is the modeled EF sampled at Alert. A 10-day boxcar smoothing has been applied to all the modeled time series between September and May. The horizontal dotted line represents the value EF=5, assumed for the emitted blowing snow SSA. (c) Surface snow salinity due to the cumulative SSA deposition on sea ice for February 1-April 30 assuming a 1cm snow accumulation rate for that period (see text). The red line shows the location of MYI.**

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